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Yordin D. Ocampo-Acuña, Enrique Salazar-Rios, M. Ángeles Ramírez-Cisneros, Maria Yolanda Rios

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Comprehensive review of liquid chromatography methods for fumonisin determination, a 2006-2022 update

Yordin D. Ocampo-Acuña,¹ Enrique Salazar-Rios,² M. Ángeles Ramírez-Cisneros^{*1} and Maria Yolanda Rios^{*1}

¹Centro de Investigaciones Químicas, IICBA, Universidad Autónoma del Estado de Morelos, Avenida Universidad 1001, Col. Chamilpa, Cuernavaca 62209, Morelos, México
²Instituto Mexicano del Seguro Social, Hospital de Especialidades del Centro Médico Nacional Siglo XXI, Servicio de Gastrocirugía. Ciudad de México, México.

Corresponding authors:

M. Ángeles Ramírez-Cisneros – angelesrc@uaem.mx Maria Yolanda Rios – myolanda@uaem.mx

¹Centro de Investigaciones Químicas, IICBA, Universidad Autónoma del Estado de Morelos, Avenida Universidad 1001, Col. Chamilpa, Cuernavaca 62209, Morelos, México

ORCID:

Yordin D. Ocampo-Acuña (0000-0001-9349-465X)

Enrique Salazar-Rios (0000-0002-0054-322X)

M. Ángeles Ramírez-Cisneros (0000-0003-4696-7359)

Maria Yolanda Rios (0000-0002-8875-8734)

Comprehensive review of liquid chromatography methods for fumonisin determination, a 2006-2022 update

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Abstract

Fumonisins are mycotoxins present worldwide. They are mainly found in corn and its derived foods; however, they also have an important presence in other grains, fruits, and vegetables. Their consumption in excessive amounts can affect animal and human health. The most abundant of these is fumonisin B₁, associated with a range of toxicological effects in animals, including equine leukoencephalomalacia, porcine pulmonary edema, and rodent carcinogenicity. In humans this mycotoxin has been shown to increase rates of esophageal cancer. The International Agency for Research on Cancer has classified FB₁ within the 2B group, considering it a possible human carcinogen. Thus, analytical methods that identify/quantify fumonisins become a necessity to ensure adequate control of food and crops. An analytic method needs to be sensitive, selective, and robust to provide reliable data that can aid in monitoring risk assessment, quality control, and research. Recently, colorimetric methods which use immunologic and molecular approaches based on dyes, enzymes and aptamers have gained attention; some of these using nanomaterials. However, these methods are still in development. Currently, chromatographic methods remain the most confident and robust analytic tool, especially for quantification purposes. There is a great deal of information reported in the literature regarding these methods; despite this, there has not been a compilation of the methods for fumonisin analysis to facilitate its consult since 2005. Being the most common method for fumonisin detection worldwide, the present review focuses on the compilation of liquid chromatography methods published between 2006 and 2022 organized by matrix, analytes, instrument, and method conditions, using diverse detectors including MS, fluorescence, and an evaporative light scattering detector. Additionally, These techniques have been applied to diverse matrices, namely food and beverages, including grains, milk, meat, beer, wine; as well as biological samples such as urine, plasma, serum, and tissues. Other aspects pertaining to legislation, extraction, cleanup (selective pressurized liquid extraction, strong anion-exchange, immunoaffinity chromatography, and QuEChERS), derivatization procedures, limit of detection and quantification of fumonisins are also included. This review had compiled and organized 88 chromatographic methods for fumonisins analysis, and the analysts can consult all the procedures with detail.

Keywords

Fumonisins, fumonisin B1, fumonisin analysis, food analysis, mycotoxins analysis method

Introduction

Despite the current improvement in processing, packing and labeling activities, food safety is still an important concern, not only for human consumption, but also for crop control, fresh food

quality and safety. Fungi contamination of these and other products is a paramount problem, as it can cause diverse ailments to humans and animals, as well as compromise production yield of the different crops and livestock. Mycotoxins are small secondary metabolites (molecular weight -MW- \sim 700) produced by microfungi; these are naturally occurring substances that are responsible for detrimental effects to the host, and are, for the most part, resistant to food processing (Bullerman 2007, Turner 2009). These compounds can be carcinogenic, nephrotoxic, hepatotoxic, neurotoxic, immunosuppressant, and can modify estrogen production (Jia 2014). An important aspect pertaining to the consumption of mycotoxins is their ability to accumulate within an organism. Thus, different sources such as grains: wheat (Headly 2022 in graphical abstract), oats, rice (Toro 2022 in graphical abstract), barley, and corn (Diogo 2011 in graphical abstract), fresh vegetables (Cumming 2022 in graphical abstract) and fruits (apples, raisins, and nuts) contribute to increase the amount of accumulated toxins in the host. This phenomenon continues in livestock whereby the ingestion of contaminated food sources increases the levels of toxins within their organisms, and are passed on to their derivatives (i.e. meat, milk, eggs, among others). As a result, human consumption of these products multiplies the chain of transmission, as crops and livestock (Embrenhar 2022 in graphical abstract) become saturated of mycotoxins from different sources; this is known as a carryover effect (Marasas 2001). Hence, contamination by mycotoxins has been recognized as a health problem, with special attention being put on aflatoxins, ochratoxins and fumonisins by their direct or accumulated toxicity (Requena 2005). Mycotoxins are generally characteristic to a specific genus. Some of the main genus producing mycotoxins are Aspergillus (aflatoxins and ochratoxins), Penicillium (patulin, ochratoxin A, citrinin, penicillic acid, cyclopiazonic acid, and penitrem), and Fusarium (trichothecenes and fumonisins) (Grajewski 2012). Among aflatoxins, ochratoxins, and fumonisins, these last ones have been associated with important human diseases such as esophageal cancer (Marasas 2001), with an increased incidence of human immunodeficiency virus (HIV) infection (Williams 2010), liver and kidney disease, and growth impairment (Chen 2018). Some reviews have compiled the toxicity and mechanism of action of FB (Chen 2021, Stockmann-Juvala 2008). It has been estimated that mycotoxins are present in at least a guarter of the world's agricultural products. and their stability at high temperatures guarantee their integrity even after passing through cooking and industrial procedures (Williams 2010). Despite these considerations, not all countries have legislation that regulate their concentration in food. The number of mycotoxins that are known to exert a toxic effect on human and animal health is constantly increasing, for this reason, generation and observance of legislation that ensures minimization of mycotoxins exposure is needed to ensure the quality of food (Bueno 2015). Diverse detection methods have been used to evaluate fumonisins, and some new methods have a promising future for easier and faster methodologies. Enzyme-linked immunosorbent assay (ELISA) methods based on antigens are specific and commercially available, however these have expiration date and need to be stored under refrigeration. Some enzymes have been proposed for colorimetric methods intended for more analytes, however these demonstrate low selectivity. Nanomaterials have arisen as a promising tool for mycotoxin detection, using immunoreactions or aptamers for detection.

Despite this, for research purposes, characterization of nanomaterials is required, and instrumentation is expensive. Thus, this method may only prove favorable for future commercial applications if a high specificity, especially in real samples, can be achieved. These techniques have been recently reviewed (Majdinasab 2021) and remain out of the scope of the present paper. In general, the most extensively used technique for mycotoxin determination is liquid chromatography associated with different detectors (Bueno 2015). This is because it has a well established and robust methodology that has been proven for all kinds of matrices. There is a considerable number of articles regarding fumonisin analysis (including reviews); however, there has been no compilation of this information available since 2006. This review aims to compile and organize the advances in the field from 2006-2022 in a single document including liquid chromatography (UPLC) methods currently used. Additionally, matrices, pretreatment procedures and instrument conditions are also reported, so that readers can easily find a method close to their needs in a single article.

1 Fusarium genus

Fusarium genus (syn Giberella) was first described by Link in 1803. It belongs to the Nectriaceae family and is widely spread in soil. Fusarium includes more than 150 species of filamentous fungi, classified into nine categories, and is considered one of the most mycotoxigenic genus. Fusarium phylogeny and morphology has been recently reviewed generating an online identification database (Crous 2021). It is of agricultural concern for its capacity to grow on plants, particularly crops, but also in fruits, contaminating food and feed (Tapia 2014, Grajewski 2012). Approximately 20 species are considered pathogenic for their capacity to produce mycotoxins that affect plants, animals, and humans. F. verticillioides and F. proliferatum are the main producers of fumonisins (Gelderblom 1988); F. solani and F. oxysporum have been reported to cause minor health problems directly to humans, producing keratitis, endophthalmitis, onychomycosis, cutaneous and subcutaneous infections, sinusitis, arthritis and mycetoma. In immunocompromised patients, however, especially those with hematological disorders, they can cause severe disseminated infections that can reach mortalities of almost 100% ("Fungal Infections. Fusarium Solani" https://www.life-worldwide.org/fungaldiseases/fusarium-solani; "Fungal Infections. Fusarium Oxysporum," https://www.lifeworldwide.org/fungal-diseases/fusarium-oxysporum). Prolonged exposition to these fungi can also lead to chronic diseases such as cancer (Shier 2000). The distribution of Fusarium species has been studied mainly in commercial substrates, and particularly for certain geographical areas such as F. graminearum and F. culmorum in Europe (Pasquali 2016), F. oxysporum in Israel and Middle East (Maymon 2020), and F. oxysporum worldwide (Dita 2018).

2 Fumonisins

The first report regarding fumonisins was published in 1988 when they were first isolated by Gelderblom *et al* (Gelderblom 1988). The chemical structure of these mycotoxins was first

proposed in the same year as a result of the collaboration between the Programme on Mycotoxins and Experimental Carcinogenesis (PROMEC) and the Council for Scientific and Industrial Research (CSIR) (Marasas 2001). Structurally, fumonisins are characterized by a long chain of polyhydroxy alkylamines containing two propane tricarboxylic acid moieties (tricarballylic acid, TCA) that are esterified to hydroxyl groups on adjacent carbon atoms. Currently twenty-eight different structures of fumonisins have been described (Agriopoulou 2020), which have been classified into four series: Series-A corresponds to amides, Series-B exhibits a free amine group and a terminal methyl, Series-C includes a terminal amine group, and Series-P incorporate an 3-hydroxypiridinium residue in their structures (Yazar 2008, Braun 2018).The fumonisins most frequently isolated from *Fusarium* are illustrated in **Figure 1**.

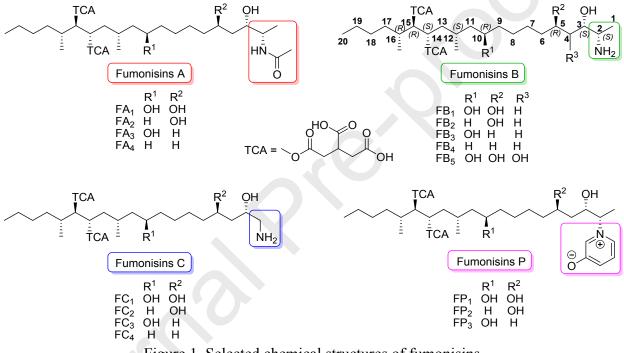


Figure 1. Selected chemical structures of fumonisins.

Within these groups of natural compounds, fumonisins B (FB₁, FB₂, FB₃) are the most relevant because they have been found on various food products and crops (Arranz 2004). FB₁ is the most abundant and toxic fumonisin of the group. Its chemical structure is a 2*S*-amino-12*S*,16*R*-dimethyl-3*S*,5*R*,10*R*,14*S*,15*R*-pentahydroxyeicosane, in which hydroxyl groups at C-14 and C-15 are substituted with a propane-1,2,3-tricarboxylic acid (TCA) residue. FB₂ does not have the hydroxyl group at C-10. FB₂ and FB₃'s structural isomers, differ only in the location of an hydroxyl group (Figure 1) (Bryła 2013). The *FUM* genes have been identified as the responsible for fumonisin biosynthesis (Alexander 2009).

2.1 Fumonisins in food

Funonisins are present in a wide number of food products around the world. Cereals are the group with the highest documented concentration of these toxins (Kamle 2019). Maize, and

maize-based products are particularly affected (Stępień 2011), with as much as an estimated 50% of products contaminated in varying degrees (Pagliuca 2005), depending mainly on agroclimatic and storage conditions (Bryła 2013). In particular, FB₁ has been found in different types of food such as asparagus, garlic (Seefelder 2002), barley (Park 2002), beers (Kawashima 2007), dried figs (Heperkan 2012), and milk (Gazzotti 2009). Additionally, FB₁ and FB₂ have been reported in 'black oats' feed from Brazil, and forage grass in New Zealand. They have also been found in home-grown corn consumed in rural areas of Southern Africa, and in commercial corn-based human food products from retail outlets (Norhasima 2009).

Concentrations of FB₁ and FB₂ vary widely between products. They have been found in corn meal up to 2.98 μ g FB₁/g and 0.92 μ g FB₂/g, and in corn grits up to 2.55 μ g FB₁/g and 1.07 μ g FB₂/g, respectively. In contrast, Switzerland, the United States, and South Africa have reported very low concentrations of these toxins, being lower than 0.06 μ g/g, in products such as corn breakfast cereal (Norhasima 2009). A meta-analysis including contamination of cereal-based foods revealed the highest concentration of fumonisins in corn-based products, followed by wheat-based products, other cereals, and barley-based foods. Regarding the occurrence, it was reported widely in other cereal-based foods, followed by corn-based foods, rice-based foods, and wheat-based foods (Farhadi 2021).

2.2 Stability

The integrity of fumonisins depend on a combination of conditions that include temperature, pH, humidity, biotic or abiotic conditions, matrix and, time in these conditions. Several studies on fumonisin stability were performed in the 90's. It has been shown that FB_1 is partially hydrolyzed at acidic or basic conditions, or at 100-125 °C, and completely degraded at 200°C for 60 minutes in the absence of a matrix (Jackson 1996). Thus, the extent of FBs degradation, and their toxicity in food depend primarily on the cooking and processing conditions (Humpf 2004). FBs are known to be relatively heat stable and are minimally affected during food processing techniques such as baking, frying, broiling or extrusion cooking, where temperatures can reach 150-200°C (Humpf 2004). In maize flour, at neutral and acidic conditions, FBs were reported stable at temperatures greater than 220°C (25 min) (Bryła 2017). Selection and disposal of damaged grains, along with soaking and/or washing corn reduced the concentration of FBs by eliminating it from food material (Saunders 2001). Dry milling has been shown to maintain FB₁ mostly intact (Kamle 2019), however, wet milling has been shown to produce products suitable for animal and human consumption (gluten, fiber, germ, and starch), as the water used in the process causes FB₁ deterioration (Saunders 2001). Fumonisins can also interact with aminoacids, proteins or reducing sugars to form covalent bonds during heat processes. For instance, FB₁ reacts with D-glucose, present in corn grits, during extrusion cooking at 160-180 °C and forms the reaction product N-(carboxymethyl) fumonisin B₁ known as NCM (Seefelder 2002, Taylor 2012).

2.3 Toxicological effects

Fumonisin has been proven to induce growth and lipid disruption in plants, animals, and humans, especially FB₁. Additionally, immunotoxicity, organ toxicity (liver, kidney, intestinal tract, heart,

lungs, brain) and reproductive toxicity has been reported (Chen 2021). Structural similarity between sphingosine, sphinganine and fumonisin (*e.g.* FB₁, **Figure 2**) is cited as the key for their toxic effects, however oxidative stress, endoplasmic reticulum stress and altered tumor necrosis factor (TNF) signaling pathway, has also been recognized as a mechanisms of their toxicity (Chen 2021; Stockmann-Juvala 2008).

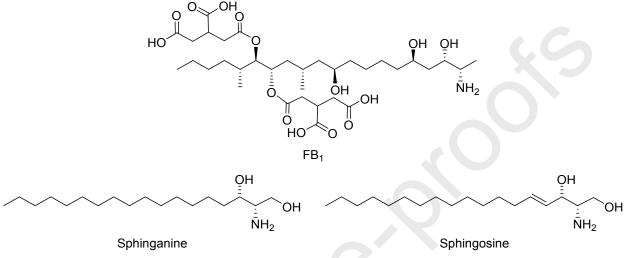


Figure 2. Chemical structures of sphinganine, sphingosine and FB₁.

In banana plants, FB_1 decreases the activity of certain enzymes such phenylalanine ammonia lyase (PAL), β -1,3-glucanase (GLU), and chitinase (CHI). It also enhances reactive oxygen species like malondialdehyde (MDA) and hydrogen peroxide, as well as transcription of genes associated to cell death (Xie 2021). In maize, FB_1 competitively inhibits ceramide synthetase (CerS) disturbing lipid equilibrium and cell protection (Beccaccioli 2021).

In animals, the presence of FBs has been found to impair immune function, cause liver and kidney damage, decrease weight and increase mortality rate (Akande 2006). Fumonisins can cause an ample range of animal diseases, including leukoencephalomalacia (LEM) in horses (Lockett 2021 in graphical abstract) and rabbits, hemorrhage in rabbits, pulmonary edema in pigs, and liver cancer in rats. In addition, they are toxic to turkey poults and have been associated with diarrhea and reduced body weight in broiler chicks (Ghiasian 2009). Different species of fish are affected by FB₁, in general, they induce weight and hematocrit reduction, as well as liver and kidney damage similar to other animal species (Oliveira 2020).

Fumonisins are associated with an increased risk of esophageal and liver cancer in humans (Liu 2017), and with a general increase of cancer incidence in regions where maize is the population's dietary base (Martins 2012). The inhibition of CerS causes the accumulation of the sphingoid bases sphinganine (Sa) and sphingosine (So), and a decrease of complex sphingolipids (Cano-Sancho 2012). Currently, the interference with sphingolipid biosynthesis remains the main cause of toxicity in humans and animals (Soriano del Castillo 2007). Sphingolipids have recently been associated with control of cell growth and proliferation of cancer cells. Ceramide has an important role in limiting cancer progression by inducing cell death (Ogretmen and Avenue 2018). Thus, its inhibition by fumonisins can potentially enhance the development of cancer,

which is why the International Agency for Research on Cancer (IARC) has classified FB₁ as a probable carcinogenic to humans (group 2B) (Duarte-Vogel 2006). Exposure to fumonisins has also been shown to increase the risk of neural tube defects (NTD) in humans (Seyed Amir Ghiasian 2006). Furthermore, some studies have suggested a possible link between exposure to fumonisins and an increase in the mortality of infection by human immunodeficiency virus (HIV) in sub-Saharan Africa (Williams 2010). More recently, a preliminary study has demonstrated the presence of hydrolyzed FB₁ (aminopentol) in the urine of women infected with human papillomavirus (HPV) and its absence in healthy women (Ramírez-Cisneros 2020). *Fusarium* produces fumonisin to facilitate its entrance to the cell by producing lipid disruption in the host cell. As a corollary, cells affected by fumonisins become a target for other infection agents such as viruses. Additionally, this lipid disruption leads to alterations in cell metabolism that can lead to cancer and cell death.

Hydrolyzed fumonisins are structurally more similar to Sa and So, however their toxic effects are still unknown. Toxicodynamic studies, especially in humans are necessary to establish dose-response of fumonisins and their hydrolyzed forms.

2.4 Toxicokinetics

The bioavailability, distribution, and toxicokinetic studies in several animal species including laboratory rodents, primates, swine, ruminants, and poultry have shown that fumonisins are poorly absorbed and have a very low bioavailability. However, little amounts of fumonisins accumulate in tissues and organs (Shier 2000). The bioavailability for FB₁ administered orally in non-human primates has been reported as < 5 % of the dose with $T_{max} = 1.02$ h. Elimination half-live was found to be $T_{1/2} = 3.15$ h for plasma, $T_{1/2} = 4.07$ h for liver and $T_{1/2} = 7.07$ h for kidney. In contrast, when administered with feed, concentrations in the kidneys increase approximately 10-fold compared to liver concentrations; suggesting an increase in the rate of elimination (Voss 2017). Bioavailability studies have demonstrated that, of the total concentration of FBs (FB₁₊₂₊₃) in the liver or kidney of rats, FB₁ shows the highest concentration, finding FB₂ and FB₃ in very minor concentrations (Voss 2017). In contrast, FB₁ is only detected in plasma and tissues at low levels, suggesting that its absorption is negligible.

Indeed, in cows and laying hens, systemic absorption of orally given FB₁ is less than 1% (Bouhet 2007). Fumonisins were mostly excreted, almost unchanged, in feces and only a small percentage was excreted in urine. Nevertheless, urine is the most acceptable, and easiest, medium to investigate compared to feces (Van Der Westhuizen 2013).

Even though fumonisins have poor absorption, they have been demonstrated to be an important factor in the development of livestock and human diseases (Shier 2000). This poses the interesting question of why they have proven toxic effects despite their low bioavailability. Several investigations have tried to explain this phenomenon, including *in vitro* studies using Caco-2-cells to prove the absorption of FB₁ in enterocytes. A study has established that the only form readily absorbed corresponds to the completely hydrolyzed form of FB₁ (aminopentol). Another study using radiolabeled FB₁, performed in nonhuman primates, demonstrated that after 24 hours of administration, the intestinal epithelial cells contained 25% of the dose (Shephard 1992). Furthermore, recent data has indicated an interaction between FB₁ and cholesterol and/or

bile salts, which may lead to the incorporation of FB_1 into mixed micelles. Thus, the metabolism of fumonisins could lead to an increased bioavailability (Bouhet 2007). Some aspects of fumonisin toxicokinetics remain unknown, however, and pigs have been suggested as a model because of its similarity with fumonisin metabolism in humans (Schelstraete 2020).

3 Limits and Legislation

Removal of mycotoxins from food products has proven to be a difficult process; therefore, maximum acceptable levels have been established for human consumption to ensure the safety of these products. Guidelines have been published in response to this need, that dictate the maximum concentration of these compounds that can be tolerated. There is a varied range of permissible amounts of mycotoxins in food according to different guidelines, encompassing ranges from 200 to 4000 µg/kg (Ponce-García 2018). Many organizations worldwide oversee strict regulations for mycotoxin control, and possible food contamination. Some of these are global organizations such as the Joint Expert Committee on Food Additives (JECFA); the scientific advisory board of the World Health Organization (WHO) and the Food and Agriculture Organization (FAO). Others are limited to geographical areas such as the European Food Safety Authority (EFSA) in the European Union, which gives counseling to European Commission; and the Food and Drug Administration (FDA) in the United States of America (Pereira 2014). In 1997, fumonisins as a subgroup of mycotoxins, were subject to regulations in only one country (FAO, 1997). In 2005, the number of countries regulating fumonisins increased to six, and the limit for their presence in maize was established as a maximum of 3000 mg/kg (Panel 2015).

Currently, many countries have implemented several regulations to control the presence of fumonisins in food products by implementing prescribed acceptable and maximum limits (WHO-Department of Food Safety and Zoonoses 2018). The JECFA established a maximum tolerable daily intake (PMTDI) of 2 µg/kg b.w./day for FB₁, FB₂, and FB₃ (alone or in combination). On the other hand, the European Union (EU Regulation 1126/2007) and the US, proposed acceptable upper limits of 4000 μ g/kg for FB₁ and FB₂ (Agriopoulou 2020). These established safe limits are not homogenous as different countries change them mainly in relationship to food products. For example, the maximum permissible levels (MPL) for the combination of FB₁ + FB₂ is 4000 µg/kg for unprocessed maize; whereas for maize intended for direct human consumption is 1000 µg/kg; 800 µg/kg for maize-based breakfast cereals/snacks; and 1400/2000 µg/kg for maize milling fractions of particle size greater/less than 500 µm respectively. The Codex Alimentarius Commission on Food Contaminants recommends a limit of 5000 µg/kg for combined FB₁ + FB₂ + FB₃ MPL for unprocessed corn grain and 2000 µg/kg MPL for processed maize-based products including flour (Bryła 2013) (WHO-Department of Food Safety and Zoonoses 2018). The main purpose of these legislations is to prevent the consumption of food that is potentially contaminated with mycotoxins, ensuring the protection of the inhabitants of developed countries (Alberts 2017). At present, there are limits established for raw maize (4000 µg/kg), as well as for maize flour and semolina (2000 µg/kg) (Alimentarius 2019). The European Commission has regulated acceptable levels of fumonisins with its most

recent modification in 2010 indicating 2000 μ g/kg for raw maize, 1000 μ g/kg for maize products for coction, 400 μ g/kg for direct ingest maize products and, 200 μ g/kg for babies and kinder food (European Comission 2007). In contrast, countries with emerging economies lack similar regulations or have poor standards; this can lead to problems with overconsumption of food with high levels of mycotoxins, including fumonisins (Ponce-García 2018). To control and/or verify fumonisin presence in food and feed products, analytical methods are needed for a wide variety of matrices. These have been proven to affect fumonisin stability and thus, bioavailability (**Tables 1-3**).

4 Analytical methods

There are a lot of reported methods for fumonisin analysis. These have been mainly developed to analyze their presence in grains and grain-based products as there is a high concern for their presence in these types of matrices. However, other matrices such as fruits, vegetables, animal tissues, cereals and beverages should also be considered, as their carry over and cumulative effects ensure their presence in these types of food products. Moreover, analysis in human matrices is of special importance to completely establish toxicokinetics, as well as to elucidate the mechanisms by which fumonisins relate to some diseases.

This review compiles and organizes 88 analytical methods for fumonisins between 2006-2022, including liquid chromatography coupled with MS detectors (single quadrupole -sQ-, triple quadrupole -QQQ- and time of flight -TOF-, with or without ion tramp), fluorescence and light scattering. The workflow for fumonisin determination includes 1) extraction, sometimes followed by 2) clean up or derivatization, and finally 3) separation and detection (**Figure 3**), being the first and third steps the fundamental ones. The detailed methodology used depends on the matrix analyzed, as well as the instrumentation available (Ridgway 2012). Matrices included in this work were classified as maize and corn-based products (34 methods), other cereal and seeds (11 methods), beverages (12 methods), products of animal origin (17 methods) and other samples (14 methods). Instrumentation used and conditions are detailed. **Table 1** includes methods describing extraction and separation/detection using chromatography coupled to mass detectors; **Table 3** refers to methods describing extraction and separation/detection using chromatography coupled to mass detectors; **Table 3** refers to methods describing extraction and separation/detection using chromatography coupled to mass detectors; **Table 3** refers to methods describing extraction and separation/detection using chromatography coupled to mass detectors; **Table 3** refers to methods describing extraction and separation/detection using chromatography coupled to mass detectors.

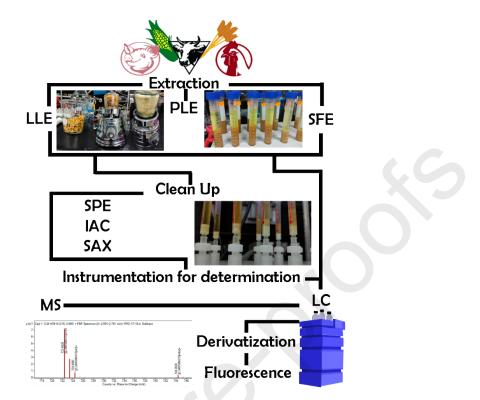


Figure 3. Workflow for fumonisin chromatographic analysis.

4.1 Extraction

Extraction is needed to obtain the enriched extract with the desired analytes, and to enhance sensitivity of the method, diminishing interferences with other components of the sample. Organic solvents, such as chloroform and hexane, which are commonly used in other mycotoxin extraction, are not recommended for FBs determination (Patel 2011); this is due to the structure of FBs, which includes multiple hydroxy, amine and carbonyl groups that make polar solvents necessary for its extraction (Scott 1993). Therefore, a mixture of water and acetonitrile (ACN) or methanol (MeOH) is the most used solvent. However, some matrices are aqueous rendering these mixtures useless as the matrices are miscible with these solvents.

FBs' ability to conjugate with proteins and sugars, allows it to be extracted with organic acids, the most commonly used are acetic acid (AcOH), formic acid (FA) and trifluoracetic acid (TFA); some authors have even used strong acids such as hydrochloric or sulphuric acid in the extraction of FBs (Zöllner 2006). To enhance the solubility of fumonisins in organic solvents, pressure has sometimes been used during extraction. Reported methods include liquid-liquid extraction (LLE) (Lucci 2015), pressurized liquid extraction (PLE) (D'Arco 2008) and supercritical fluid extraction method to be used and/or the extraction yield (Damiani 2019).

Reported methods use an aqueous:organic proportion ranging from 10 to 85 % of organic solvent, however, typically more than 50% of MeOH or ACN and, from 0.1 to 3% of acid is used. Some mixes of ACN:MeOH:H₂O were used keeping the mentioned range for aqueous, and some used 100 % ethyl acetate (Monbaliu 2009) for extraction. Immunoaffinity extraction was

also reported for urine samples. Usually, suspension of sample into extraction solvents, was followed by shaking, for periods of time ranging from seconds up to 3h, filtering or centrifugation, from 3,000 to 10,000 rpm for 2 to 15 min (**Tables 1-3**).

In 2012 Pietri and collaborators observed problems during the extraction step which resulted in unexpected low recoveries in maize flour samples due to the interactions between fumonisins and matrix components (Damiani 2019)

4.1.1 Liquid-liquid extraction (LLE)

LLE is the most commonly used technique which, depending on the composition of the food matrix, uses a mixture of acidified solvents (Lucci 2015). Examples include: methanol-water (Paepens 2005), acetonitrile-water (Zitomer 2008) or methanol-acetonitrile-water and a non-polar phase (Bryła 2013). It is based on the distribution of toxin in immiscible phases (aqueous and organic phase). The non-polar contaminants (lipids and cholesterol) are removed with non-polar organic solvents such as hexane and cyclohexane, while polar toxin compounds are extracted in the aqueous phase. This method is useful for both liquid and solid samples, the latter are homogenized and remain suspended in a polar solvent. In both cases, centrifugation is carried out, after which drying is performed under a nitrogen atmosphere, and, finally, reconstitution is done in a mixture of the chosen solvent. LLE is suitable for several toxins at small-scale preparations, however, its main disadvantage is that it is time consuming and there can be loss of sample during handling (Nawaz 2017).

4.1.2 Pressurized liquid extraction (PLE)

PLE, also known as Accelerated Solvent Extraction (ASE), uses temperatures around 100-180 °C and 1500-2000 psi of pressure to modify the conditions of the solvent and the sample, and facilitate the extraction of analytes (Kou 2003). The sample is initially dispersed with an inert material and further loaded into an extraction cell where the solvent is pumped in. Then, the extraction cell is heated to the desired temperature (above 200 °C) for 5 to 9 minutes, and pressurized (D'Arco 2008).

4.1.3 Supercritical fluid extraction (SFE)

Supercritical fluids are helpful in the extraction of analytes from a matrix. Their unique properties (low density and viscosity) make them superior to conventional extraction solvents, facilitating the extraction of compounds in samples. The most used fluid is CO₂, however, analytes with polar characteristics do not adequately dissolve. To increase its efficiency towards polar analytes, modifiers such as methanol, ethanol or acetone are added. Limitations of this technique include high cost and the need for sophisticated equipment (Selim 1996, Nawaz 2017).

4.2 Clean-up

A great number of methods have included a clean-up step after extraction (**Table 2**). The aim is to eliminate major impurities like organic acids, polar pigments, sugars, among others. The most used are QuEChERS, solid-phase extraction (SPE) with reverse phase, strong anion exchange (SAX) cartridges, and immunoaffinity columns (IAC) (Damiani 2019, Marschik 2013). It has

been shown that solvent temperature used in this process can deeply influence the recovery of fumonisins (Lawrence 2000).

4.2.1 QuEChERS

QuEChERS is a technique initially developed by Anastassiades and collaborators in 2003 (Anastassiades 2003). They coined the acronym QuEChERS which stands for Quick, Easy, Cheap, Effective, Rugged and Safe. It involves micro-scale extraction with acetonitrile, followed by a cleanup based on a dispersive solid-phase extraction (d-SPE) (Wilkowska 2011). In the extraction step, magnesium sulphate is used to reduce water in the sample, along with sodium chloride, while in the cleanup step, a primary secondary amine (PSA) or C_{18} is usually used as sorbent to retain co-extracted compounds such as sugar and fatty acids (Ridgway 2012, Zhang 2012). Other salts such as magnesium chloride, sodium nitrate, sodium sulfate and lithium chloride have been used to eliminate water, finding magnesium sulfate as the most effective for separation of both phases, eliminating water from the organic phase. QuEChERS has become the most popular pre-treatment for some matrices like as corn, wheat, oats, rice, and other cereals, as it boasts several advantages such as the decrease in volume of solvent, materials, time, as well as a reduction in cost of analysis.

4.2.2 Solid phase extraction (SPE)

SPE is a variation of traditional chromatography, and thus, is based on the same principle, the use of a mobile and a stationary phase. Separation is performed according to affinity using small disposable cartridges packed with silica gel or bonded phases which are in the stationary phase. The sample is first dissolved and loaded into a cartridge, after which it is rinsed to remove most of the contaminants and is subsequently extracted from the cartridge with a polarity compatible solvent. All this is done under reduced pressure. The SPE cartridges contain different binding phases, for example silica gel, C₁₈ (octadecylsilane), floredil, phenyl, aminopropyl, ion exchange (anionic and cationic) or SAX, immunosorbents, and molecular imprinting polymers. These last two are affinity materials which provide them with a high binding capacity for small molecules making them excellent candidates for cleanup in terms of specificity, however, they have a high cost, and are not compatible with organic solvents, limiting their use to aqueous systems. This is a disadvantage compared to more common binding phases such as SAX or C₁₈ (Turner 2009). Regarding fumonisin analysis, C_{18} is the most used stationary phase for SPE due to its easy acquisition, low costs, and the possibility of extraction of hydrolyzed forms. The second most used phase are SAX resins, whose efficiency is based on the interaction with fumonisin carbonyl groups, making them not appropriate for hydrolyzed forms (Zöllner 2006). Its elution has been reported with MeOH acidified with 0.05% AcOH achieving a pH < 7. When ion exchange resins are used for this purpose, it is necessary that the analyzed mycotoxin be in its ionic form and in an aqueous solvent. For this reason, pH regulation of the medium is an important factor. This methodology has been used for the extraction of fumonisins and moniliformin. SAX columns consist of resins with weakly basic functional groups, such as NH₂, NHCH₃ or N(CH₃)₂, or with quaternary ammonium strongly basic groups (N(CH₃)OH) in which OH is replaceable by

mycotoxin. Several types exist in both, anionic and cationic phases. SAX is the favored material for mycotoxin extraction (Turner 2009).

IAC uses antibodies, present in the stationary phase, that bind selectively to mycotoxins present in the extract. This poses an important advantage, as there is a specific interaction between the antibody and the analyte, resulting in a greater speed of interaction. After antibody binding, mycotoxins are recovered by elution with a miscible solvent or by antibody denaturation. Disadvantages of this process include the necessity of combination with other techniques such as LLE or SPE for complex samples; and the requirement for the extract to be in aqueous solution containing little or no organic solvents, as their presence, even in low concentrations, can denature antibodies (Pereira 2014). Recently, a rapid and sensitive method for determination of seven mycotoxins (including FB₁) using immunomagnetic (monoclonal antibodies conjugated with CNBr) solid-phase extraction (IMPSE) coupled to UPLC-MS/MS has been developed for peanut, maize, and wheat matrices (Wang 2022).

4.3 Derivatization

The main objective of derivatization is to change the chemical and physical properties of compounds by modifying their chemical structure (Qi 2014). Thus, derivatization reagents react with target compounds containing various functional groups, including carbonyl (O'Brien-Coker 2001), hydroxyl (Barry 2003), carboxyl (Santa 2009), amine (Vanhoenacker 2009), and thiol (Vichi 2013).

This strategy has been of utmost importance in the development of new methodology for the detection of fumonisins, as these compounds are not capable of developing fluorescence or absorbance in UV-VIS light, due to their lack of a suitable chromophore or fluorophore group for detection. Derivatization with fluorescent derivatives including 9-fluorenylmethylchloroformate (FMOC-CL), 4-flouro-7-nitro-benzofurazan (NBD-F), o-phthaldialdehyde (OPA), naphthalene-2,3-dicarboxaldehyde (NDA) and dansyl chloride (DnS-Cl) (Ndube 2011, Silva 2009) allows for fumonisin detection with HPLC coupled to fluorescence or UV-VIS 2009), albeit with a low sensitivity. Despite these limitations, UV detection is still used although the methods are not new (Cardinael 2015). Out of the fluorescent derivatives, OPA is the most used due to its low detection limits (50 ng/g), followed by NBD-F which is detected at 100 ng/g. NDA has an even lower detection limit than OPA, however, its use is generally avoided as potassium cyanide is required during derivatization, representing a high health risk (**Table 3**).

4.4 Instrumentation for determination

Once the sample is obtained, extracted and, in some cases purified or cleaned-up, different instrumentation can be used for fumonisin analysis; being HPLC and UPLC the most frequently employed. Chromatographic column is used in a reverse phase, most commonly with C_{18} as a stationary phase; nevertheless, diphenyl, amide and C_8 may also be used. The chromatographer can be coupled with a fluorescence (**Table 3**) or ESI source with mass spectrometry detectors (**Tables 1-2**). For these last ones, QQQ is the most widely used analyzer, although sQ and TOF analyzers had also been utilized. These are all used in positive mode and all acquisition modes

are reported, full scan, single reaction mode (SRM) or multiple reaction mode (MRM). Some analyzers use an Ion Trap array.

4.4.1 Separation

Fumonisins have a higher molecular weight (around 721.83 g/mol) compared to other mycotoxins such as Ocratoxin A (403.81 g/mol), Zaralenone (318.36 g/mol) or Patulin (154.12 g/mol). Because of their high polarity, reverse phase LC is an excellent option for its separation. Previous extraction methods involve an aqueous phase, which is included as mobile phase (**Tables 1-3**). Different proportions of solvents are used for the composition of the mobile phase, MeOH:H₂O is preferred, followed by ACN:H₂O, especially when derivatization is used to provide better sensibility (Velázquez 2000). There is a clear tendency of using a greater proportion of organic solvents in these mixtures with gradients reaching 100 % organic concentration, as well as the addition of FA or AcOH, and in some cases ammonium salts. This is done to enhance the ionization process necessary for mass detection, to control pH, and to increase the efficiency of separation.

Temperature used for these analyses vary between 10-45°C for MeOH as mobile phase and 30-50°C for ACN; flows from 0.1 to 1 mL/min are reported. Column dimension is another important aspect to consider when analyzing fumonisins. According to the literature compiled in the present article, there is a great variation between column dimensions, ranging from 50 to 250 mm in length, diameters going from 2.0 to 4.6 mm, and particle size ranging from 1.6 to 5 μ m. The most used, however, oscillate between 100-150 x 2 mm, with a particle size of 4.6 μ m. A recent work by Sultan et. al. evaluated the efficacy of 5 columns with different dimensions and particle sizes. FB₁ and FB₂ were analyzed using liquid-liquid extraction, followed by a cleaning procedure using SPE, and fluorescence detection, using OPA as a fluorophore group. They conclude that the use of reverse phase SPE, followed by derivatization with OPA is an effective method for the determination of fumonisins, which agrees with the information gathered by this review. In that work the comparison between columns Nucleosil Cronus (150 mm x 4.6 mm, 5 μ m) and Poroshell (75 mm x 4.6 m, 2.7 μ m) yielded similar results regarding time and solvent use. However, it is of note that the use of columns with porous particles or those with a solid nucleus affect separation. Similarly, both the diameter of the column, and particle size used are also important parameters to determine in fumonisin analysis (Sultan 2022).

4.1.2 Detection

Although many methods for fumonisin detection exist, such those based in fluorescence, the methods based on MS are the most sensitive. Among the methods included in this work, the lowest FB₁ LOD for fluorescence detector was 0.025 versus 0.0005 μ g/Kg obtained with MS QTrap detector. Besides, MS detectors offer a great advantage as they do not require derivatization (**Tables 1-2**).

FDA methods depend on the presence of a chromophore or fluorophore that allows for the correct detection of the analytes, as has been mentioned previously, various derivatizing agents exist (see section 4.3), despite this disadvantage, these methos are still commonly used due to their low costs, and their applicability to a great number of matrices including beer, maize, and

biological fluids, among others. Fumonisins can be detected at the following longitudes: λex : 420 nm, λem : 500 nm.

On the other hand, mass spectrometry for the detection of fumonisins is carried out with an ESI interphase, and IT, orbitrap, QQQ and TOF analyzers used in positive mode. In this analysis, the ion $[M + H]^+$ has been found to be the most abundant, with or without a high grade of fragmentation. Additionally, in negative mode, the formation of doubly charged molecular ions has been reported. The positive mode is used more frequently, although some authors have reported that it favors the formation of adducts that may present a problem with sensitivity. Despite this, the positive mode is still the most used mode as the $[M + H]^+$ ion is three times more abundant than $[M - H]^-$.

According to the present review, various mass analyzers such as sQ, QQQ, and TOF have been used, some of them with an ion trap (IT). IT methods are theoretically more sensitive, yet, not all ion trap (IT), Trap or QTrap methods reported here have been the most sensitive, with some QQQ or even sQ methods being able to detect lower concentrations (**Tables 1-2**).

Lower limits for FB₁, FB₂, and FB₃ have been reported by different authors, including Šarkanj *et al* for urine analysis (0.001 µg/L FB₁ and FB₂) (Šarkanj 2018), Zitomer *et al* regarding maize tissues analysis (0.01 µg/kg for FB₁ and FB₂) (Zitomer 2008), Huang *et al* for liquorice (0.05 µg/L FB₁ and FB₂) (Huang 2018). Among the different fumonisins analyzed, the most reported is FB₁, with [M+H] 722.2 *m/z* being the most abundant ion. Additionally, the 334.3 and 352.3 *m/z* product ions can also be obtained by using a collision energy of 38-56 and 38-40 eV respectively (**Table 4**).

A light scattering method has also been reported. Even though its reported LOD and LOQ are high, these limits approach those that are permissible. Thus, it may prove useful in screening, as detection by this method correlates with level above the permissible limits (Ramalho 2022, Mirón-Mérida 2021).

4.1.3 Non chromatographic methods for fumonisins detection

Aside from conventional chromatographic methods, there is a wide variety of methods for fumonisin determination. These can be classified into two groups: immunological and molecular (**Table 5**) (Deepa 2019).

The immunological methods are based on the interaction between the mycotoxin and a specific antibody. These antibodies act by recognizing specific chemical groups; as such, they can recognize structural analogs. To facilitate antibody detection, a marker is added which can be radioactive, chromogenic or fluorogenic in nature. The most popular, commercial, immunological method for fumonisin detection is enzyme-linked immunosorbent assay (ELISA) (Pereira 2014). This method has been used to determine fumonisin concentration in corn and other cereals (Wang 2006); fresh and dehydrated commercial garlic (Tonti 2017); during industrial cornflakes processing (Castells 2008); and maize and gluten meal (Coronel 2016). Techniques such as time-resolved immunochromatographic assays, enzyme-linked aptamer assays, chemiluminescence immunoassays, fluorescence immunoassays, fluorescence resonance energy transfer immunoassays, and metal-enhanced fluorescence assays have been implemented in the detection of mycotoxins (Majdinasab 2021, Chauhan 2016).

Although molecular methods do not directly determine the presence of fumonisins, they are nonetheless important as they allow rapid detection of fumonisin-producing species. These DNA-based identification methods are fast, sensitive, and reliable (Deepa 2017) because they are independent of the morphology and cultivability of the fungi. Of these, PCR is the most frequently used technology for detection of mycotoxin-producing *Fusarium* species (Gong 2015). Today, aptamer-based methods are having a great impact in the detection of mycotoxins. Due to their exceptional affinity and specificity, they can be comparable to antibodies, with certain advantages such as easy nucleobase and chemical modification, and exponential self-amplification (Mirón-Mérida 2021).

Also, these methods take advantages of nanomaterials to improve LOD, cost, analysis time, reduce instrument use for final users and overall, pretreatment and manipulation of samples. However, at a research level, nanomaterials need to be characterized, requiring instrumentation that is not common. Many of these technologies are still under development, with a large amount of research proposing them for fumonisin determination. Much of this information has been compiled over the years in various review papers (Majdinasab 2021, Deepa 2019, Gong 2015, Mirón-Mérida 2021). Until these methodologies achieve the robustness of chromatographic techniques, especially for absolute quantification, the latter techniques remain the techniques of choice.

			Table 1. LC-MS methods for FBs without clean up			
Re f	FBs	S a m p l e (g)	Sample treatment	LC conditions LC conditions Column / Injection volume / Mobile Phase Flow / Analysis Time		MS cond ition s, Limi ts
	Mat rix		Extraction procedure			Mass Condi tions / Limit s
			Maize and corn-based products			
(Zi to me r et	$\begin{array}{c} B_1,\\ B_2,\\ B_3 \end{array}$	0.	1Add 2 mL ACN/H ₂ O 1:1 (5% FA); 2 Gently shaken for 3 h; 3 Centrifugate to 15000 g; 4 Filter; 5	Metachem Inertsil ODS-3, 1 Inj vol 20 µL, A) H ₂ O/ACI H ₂ O/ACN/FA 2:97:1. 70-50% B in 2 min, keep 10 min; initial co	N/FA 97:2:1, B) in 9 min, 50-100% B	QTra p CaT : 210°C
al. 20 08)	Maiz e leaf	0 1	Dilute 1:10	Flow: 0.20 mL/min	Time: $t_{an}=21$ min, $t_{Tot}=31$ min	LOD: 0.01 µg/kg all FBs
(D e Gi rol am o et al. 20 14)	B ₁ , B ₂ , PHF (B ₁ , B ₂), HF (B ₁ , B ₂) Maiz e base d prod ucts	20	1 100 mL MeOH/ACN/citrate-phosphate buffer 25:25:50; 2 Shake 1h; 3 Dilute 1:10 with MeOH/H ₂ O 80:20 with 0.5% AcOH; 4 Filter	Gemini C ₁₈ , 150 x 2.0 mm Inj vol 20 μ L, A) H ₂ O, B) Me(AcOH 40-60% B in 30 min, 60 to 40% conditions for 9	OH, both with 0.5% 6 B in 1 min; initial	Orbit rap CaV 45 V; SV 4 kV; RF Lens 75 V; ST 300 °C; CaT: 300 °C; SG 30 U; GF 10 skim

$\begin{array}{c} \operatorname{mer} V \\ 18 V \\ LOD: \\ 5 \\ \mu g/kg, \\ LOQ: \\ 100 \\ 1$

				Flow: 0,2 mL/min	time: t_{an} =30 min, t_{Tot} =40 min	10 μg/kg all FBs
(B elt rán et al. 20 09)	B ₁ , B ₂ Maiz e, kern el, dry pasta , baby food	2. 5	1 Add ACN/H ₂ O 80:20 + 0.1% AcOH, 2shake 90 min, 3centrifuge to 4000 rpm, 10 min; 4dilute 1:2 with H ₂ O, 5filter (0.22 mm nylon filter)	Acquity UPLC BEH C ₁₈ , 50 x 40°C Inj vol 20 μL, A) H ₂ O, B) MeO AmAc and 0.1% 10-90 % B in 4 min, initial co	H, both with 0.5 mM AcOH	QQQ CaV 3.5 kV; DGT 500°C ; ST 120 °C; T 40 °C; DGF 1200 L/h, CoG 4 x 10 ⁻³ mbar
				Flow: 0.3 mL/min	Time: t_{an} =4 min, t_{Tot} =7 min	LOD: 1 µg/kg, LOQ: 3.5 µg/kg
(C. Dall' Asta et al. 2008)	,	2 5	LLE 1 Add 100 mL H ₂ O/ACN/MeOH 50:25:25, 2 blend (6000 rpm/5 min); 3 take 4 mL; 4 filter; 5 dry N _{2;} 6 reconstitute 1mL in H ₂ O/ACN 1:1; 7 filter	XTerra C ₁₈ , 250 × 2.1 mn Inj vol 10 μL, A) H ₂ O, B) MeOl 0% B for 3 min, 0-45% B in 2 85% B in 15 min, keep for 10 n for 10 min	H, both with 0.1% FA min, keep 5 min, 45- nin, initial conditions	QQQ CaV 3.2 kV; CV 30 V; EV 3 V; ST 120 °C; DGT 160 °C; CGF 70 L/h;

	pr od uc ts			Flow: 0.2 mL/min	Time: t_{an} =35 min, t_{Tot} =45 min	$\begin{array}{c} DGF \\ 650 \\ L/h \\ (N_2 \\ for \\ both) \\ LOD: \\ B_1, B_2 \\ 1 \\ \mu g/kg, \\ FB_3 \\ 8 \\ \mu g/kg, \\ LOQ: \\ B_1, B_2 \\ 5 \\ \mu g/kg, \\ FB_3 \\ 12 \\ \mu g/kg \end{array}$
(Arro yo- Manz anare	B ₁ , B ₂ an d ot he r to xi ns	2	QuEChERS 1Add 8 mL of H ₂ O; 2shake 10 s; 3add 10 mL 5% FA in ACN; 4shake 2 min; 5add 4 g MgSO ₄ + 1 g NaCl; 6shake 1 min; 7vortex 2 min; 8centrifuge to 4500 rpm, 5 min, 4 °C; 9take 5 mL; 10-dry under N ₂	ACQUITY HSS UPLC T3, 150 30 °C Inj vol 10 μL, A) H ₂ O, B) MeOl and 5 mM Ar 5% B, keep 0.5 min, 5-94% B min, 94-5% B in 3 min; initial	H, both with 0.3% FA nF in 19.5 min, keep 1	QQQ ST 150 °C; DGT 400 °C; NG 7 bar (N ₂); CGF 150 L/h; DGF 1000 L/h
s et al. 2018)	W he at, m ai ze		at 40 °C; 11reconstitute (0.2 mL MeOH/H ₂ O 1:1); 12centrifuge to 14000 g, 5 min, 4 °C	Flow: 0.4 mL/min	Time: t _{an} =21 min, t _{Tot} =28 min	L/H LOD; 1.28 B ₁ , 0.25 FB ₂ , 0.27 B ₃ µg/kg LOQ: 4.24 B ₁ , 0.82 FB ₂ , 0.89

	B ₁ , B ₂ , B ₃			5		B ₃ μg/kg QQQ CaV 3.2 kV; EV 3 V; ST
(C hia ra Da ll' As ta, Ga lav ern a, et al. 20 09)	Corn - base d prod	5		Xterra C ₁₈ , 250 x 2.1 mm, 5 μm, at 30 °C Inj vol 5 μL, A) H ₂ O, B) MeOH, both with 0.2% FA 30% B for 2 min, 30-45% B in 3 min, 45-90% B in 20 min, keep for 10 min, 30% B in 1 min; initial conditions for 20 min		v, 51 120 °C; DGT 160 °C; CGF 70 L/h; DGF 650 L/h (N ₂ , both)
	ucts			Flow: 0.2 mL/min	Time: t_{an} =35 min, t_{Tot} =56 min	LOD: FB ₁ 4 µg/kg, B ₂ , FB ₃ 8 µg/L OQ: B ₁ B ₂ 5, B ₃ 12 µg/kg
(C hia ra Da ll' As ta, M an gia , et al. 20 09	B ₁ , B ₂ , B ₃ Grou nd corn	5	1 Add 50 mL H ₂ O/MeOH 30:70; 2Blend to 6000 rpm, 10 min; 3 Stir for 50 min; 4 Centrifuge to 3500 g, 15 min; 5 Filter (2 mL)	Xterra C ₁₈ , 250 x 2.1 mm, 5 μm at 30°C Inj vol 10 μL, A) H ₂ O, B) MeOH, both with 0.1% FA 30 % B for 2 min, 30-45% B in 3 min, 45-90% B in 20 min, keep for 10 min; initial conditions for 15 min		Pg/Rg QQQ CaV 4 kV; EV 2 V; ST 120°C ; DGT 350 °C; CGF 50 L/h; DGF 600

				Flow: 0.2 mL/min	Time: t_{an} =35 min, t_{Tot} =50 min	LOD: 5 µg/Kg
(CB1, B2, B3DaB3DaII'II'Asta, giaCornan-giad, et al. 20ucts09))	B ₂ , B ₃ Corn - base	B ₂ , B ₃ Corn - 5 base d rod	 5 1 Add 2 ml H₂O/ACN/AcOH 20:79:1; 2 Extract 90 min in rotatory shaker; 3 Centrifuge 3000 rpm, 3 min; 4 Take aliquot 350 μL and dilute 1:1 with extraction solvents 	Gemini C_{18} , 150 x 4.6 mm Inj vol 5 µL, A) H ₂ O/ACN/ H ₂ O/ACN/AcOH 2:97:1, both 0% B for 2 min, 0-100% B in 12 initial conditions for	QQQ CaV 4.0 kV; EV 3 V; ST 550 °C; CUR 10 psi	
	-			Flow: 1.0 mL/min	Time $t_{an}=17$ min, $t_{total}=21$ min	LOD: 8 µg/kg
	$\begin{array}{c} \mathbf{B}_1, \\ \mathbf{B}_2 \end{array}$				·	QQQ CaV 4.5
(G. B. de Oli vei ra	Maiz	1	 Add 1 g Silica gel as dispersant; 2 Mix in polypropylene cartridges, MSPD; 3 Elute with 16 mL of 20 mM AmFo buffer:MeOH 9:1 (pH 7); 4 Collect 2 mL fractions; 5 Centrifuge to 4000 rpm, 10 min; 6 	Poroshell, C ₁₈ , 100 x 3 mm, 2.7 μm, 40 °C Inj vol 10 μL, A) Ultrapure H ₂ O, B) ACN, both with 0.1% FA 20-90% B in 3 min, keep 0.4 min, 90-20 % B in 0.1 min; initial conditions for 6 min		kV; EP 10 V; DGT 650 °C; NG 40 CUR 18 a.u,
et al. 20 17)	Maiz e		Filter	Flow: 0.5 mL/min	Time: t_{an} =3.4 min, t_{Tot} =9.5 min	LOD: B_1 514, B_2 176 $\mu g/kg$ LOQ: B_1 594, B_2 210 $\mu g/kg$
(D 'A rco et	$\begin{array}{c} B_1,\\ B_2,\\ B_3\end{array}$	3	1 Add 100 μL of a 5 μg/mL Fbs solution (0.5 μg) and keep 15 min at RT; 2pack into 11 mL PLE pressure resistant stainless steel extraction cell; 3elute with 22 mL of MeOH 60% at 40°C and 34 atm, 2 min of preheating, 5 min of static time, 60 s of purge time; 4concentrate to 5 mL (40 °C and 80 mbar); 5transfer to a 15 mL conical tube; 6evaporate to dryness at 55°C with N ₂ ; 7.reconstitute 1 mL MeOH/H ₂ O 50:50; 8	Luna C ₁₈ , 150x4.6 mm, 5 Inj vol NR, A) H ₂ O, B) MeOH 65% B for 3 min, 65-95% min, initial conditi	, both with 0.5% FA 6 B in 4 min, keep 3	QQQ CaV 3.20 kV;

al. 20 08)	- d baby food		filter	600		CoV 50 V; EV 3 V; RF lens 0.2 V; ST 125 °C; DGT 300 °C; DGF 500 L/h; CGF gas 50
				Flow: 0.30 mL/min	Time: $t_{an}=10$ min, $t_{Tot}=20$ min	L/h LOD: 0.7 B ₁ and B ₂ , 1.5 μg/kg B ₃ LOQ: 2 B ₁ and B ₂ , 5 μg/kg B ₃
(C hia ra Da ll' As ta, M an gia , et al. 20 09)	B ₁ , B ₂ , B ₃ Raw corn	1 0 0	 Add 50 mL KOH 2M; 2Centrifuge to 6000 rpm, 10 min; 3 Stir (50 min); 4 Add 50 mL ACN; 5 Stir 10 min; 6 Separate 20 mL and dry under N2; 7 Redissolve in 50 mL KOH 2M; 8 Centrifuge to 3500 rpm, 15 min; 9 Dry under N2; 10 Redissolve in H₂O/MeOH 30:70 	Hypersil C ₁₈ , 150 x 2.1 mr Inj vol 10 μL, A) H ₂ O, B) MeOI 20% B for 1 min, 20-100% B in min, initial condition	H, both with 0.2% FA walnut 5 min, keep 3	QQQ QTra p CaV 4 kV; CoV 50 V; ST 425°C ; DGT 350°C ; CGF 50 L/h; DGF 600 L/h (N ₂ ,

				Flow: 0.6 mL/min	Time: $t_{an}=8$ min, $t_{Tot}=13$ min	both) LOD: <15 μg/kg
(H u et al. Ra ma	B ₁ , B ₂	1	1 10 mL ACN/H ₂ O/AcOH 70:29:1; 2 Shake 30 min; 3 Centrifuge to 4500 rpm, 10 min; 4 Filter supernatant; 5 Take 1 mL; 6 Add 10 μL, 1 μg/mL ¹³ C-34 FB ₁ and ¹³ C-34 FB ₂	Luna C ₁₈ , 150 x 2 mm, Inj vol 5 μ L, A) H ₂ O, B) Met AmAc 40-90% B in 6 min, keep 1 min keep 1 min, 100-40% B in 2 min 4 min	OH, both with 2 mM , 90-100% B in 1 min,	Qtrap CaV 5.5 kV; EP 10 V; ST 600°C ; CUR 40 psi; CoV 10 V; dwell time 100 ms
				Flow: 0.2 mL/min	Time: t _{an} =9 min, t _{Tot} =15 min	LOD: 7 B ₁ ; 6 B ₂ µg/kg LOQ: 28 B ₁ ; 27 B ₂ µg/kg
	\mathbf{B}_1					QTra p CaV
(B erg ma nn, Hü bn er, an d Hu mp f 20 13)	Maiz e	1 0	1 Add 20 mL ACN/H ₂ O 70:30 with 1% FA; 2Vortex 30 s; 3 Sonicate 10 min; 4 Shake 15 min; 5 Centrifugate to 8000 g, 15 min, 25 °C; 6 Dilute 1:1 1% FA; 7 Filter if necessary	Inj vol 20 μL, A) H ₂ O, B) AC 65% B for 4 min, 37.5% B for	lyperclone C ₈ BDS, 150 x 2.0 mm, 3 μ m at 40° C nj vol 20 μ L, A) H ₂ O, B) ACN, both with 1% FA 5% B for 4 min, 37.5% B for 0.5 min, 5% B for 2 nin, keep for 0.5 min, initial conditions for 4 min	

				Flow: 0.30 mL/min	Time: t_{an} =7 min, t_{Tot} =11 min	Torr; QTrap CUR 20 psi LOD: 53 µg/kg, LOQ: 188 µg/kg
(d	B ₁ , B ₂ , HB ₁ , HB ₂			65-80% B in 3 min,	2	QQQ CaV: 3kV; DGT: 400 °C; ST: 150 °C; CGF: 15 L/h; DGF: 750 L/h
e M at os et al. 20 21)	Corn prod ucts	5	1 Add ACN:H ₂ O:FA 75.24:1; 2 shake for 2 min; 3 sonicate for 10 min; 4 centrifuge at 3000 rpm for 7 min; 5take 0.05 mL of extract; 6 dilute with 0.95 mL 0.05% of AF in MeOH:H ₂ O 1.1; 7 filter	Flow 0.3 mL/min	Time: tan= 5 min, t _{Tot} = 7 min	$\begin{array}{c} \text{LOD:} \\ (B_1: \\ 0.43- \\ 1.98, \\ FB_2 \\ 0.19- \\ 1.37, \\ HB_1 \\ 0.72- \\ 1.39, \\ HB_2 \\ 0.36- \\ 0.70) \\ \mu\text{g/K} \\ \text{g} \\ \text{LOQ:} \\ (B_1:1. \\ 43- \\ 6.59, \\ FB_2 \\ 0.60- \\ 4.60, \\ HB_12. \\ 40- \end{array}$

				2S		4.60, HB ₂ 1.20- 2.30) μg/Kg
(Li n et al. 20	B ₂	B1, B2 B2 5 1 Add 25 mL MeOH/H2O 3:1; 2Ultrasonic bath for 10 min at RT, output powder 120 W; 3 Centrifugate to 5000 g, 5 min; 4 Filter (0.22 mm nylon filter)		Zorbax Eclipse XDB-C ₁₈ , 150 x 2.1 mm, 3.5 μm at 30°C Inj vol 10 μL, MeOH/H ₂ O/FA 75:25:0.2		Q CaV 3.5 kV; CoV 50 V; ST 120 °C; DGT 350°C ; DGF 600 L/h
20 11)			Flow: 0.20 mL/min	Time $t_{an=total} = 4 \min$	LOD: 3.5 B ₁ , 2.5 µg/kg B ₂ LOQ: 11.7 B ₁ , 8.3 µg/kg B ₂	
	$\begin{array}{c} B_1,\\ B_2 \end{array}$					TOF CaV 5.5
(A. S. Sil va et al. 20 19)	Maiz e flour	2	1 Add 10 mL ACN 80%; 2 Shake at 110 rpm, 1h; 3 Centrifuge to 3000 rpm, 10 min; 4 Remove supernatant; 5 Re-extract the solid, same way; 6 Centrifuge to 3000 rpm, 10 min; 7 Dilute 1:1 with H ₂ O; 8 Filter	Zorbax Eclipse Plus C ₁₈ , 2.1 x °C Inj vol 20 μL, A) 0.1% 10-70% B in 12 min, 70-90% F 90-10% B in 1 min, initial c	5 FA, B) ACN 3 in 1 min, keep 1 min,	5.5 KV; ST 575 °C; CUR 30 psi; Gas 1 and Gas 2, 55 psi both; DP 100 V;

				Flow: 0.5 mL/min	Time: $t_{an}=14$ min, $t_{Tot}=17$ min	Full scan 100- 750 Da LOD: 62.5 μg/kg, LOQ: 125 μg/kg all FBs
			Other cereal and seeds			
(Bart ók et al. 2006)	B ₁ , B ₂ , its an al og s	3	1 Add 25 mL of ACN/H ₂ O 75:25; 2 Centrifuge to 13,500 rpm, 1 min; 3 Shake 1 h; 4 Centrifuge to 10,000 g, 10 min; 5 Filter	Supelcosil ABZ Plus, 250 x 2. Inj vol 1 μL, A) H ₂ O, B) ACN 25-40 % B in 22 min, 40-100% min.	, both with 0.1% FA	QTra p CaV 3.5 kV; EV 200 V; HED Volta ge 7 kV; NG 40 psi; DGF 9 L/min ; DGT
	Ri ce					, DG1 350 °C; trap drive 53.9; max accum ulatio n time 300 ms; full scan 50-

				Flow: 0.3 mL/min	Time: t _{an} =27 min,	1100 <i>m/z</i> LOD / LOQ:
(S ole im an y, Jin ap, an d Ab as 20 12)	B ₁ , B ₂	1 1 Add 40 mL H ₂ O/ACN/AcOH 20:79:1; 2 Shake 60 min; 3 Centrifuge the supernatant at 3000 rpm, 10 re 0 min; 4 Dilute 1:1 in H ₂ O/ACN/AcOH 79:20:1; 5 Filter	Thermo Scientific C ₁₈ , 150 x 4.6 mm, 3 μ m at 30°C Inj vol 20 μ L; A) H ₂ O, B) MeOH both with 0.1% AcOH 5% B for 8 min, 5-90% B in 14 min; 90-5% B in 3 min		NR QQQ CaV 3 kV; ST 120°C ; DGT 400 °C; spray gas N ₂	
				Flow: 0.25 mL/min	Time: t_{an} =22 min, t_{Tot} =25 min	LOD: 20 ng/g, LOQ: 40 ng/g
(Raus ch, Broc kmey er, and Schw erdtle	he r to xi ns	1	QuEChERS 1 Add 2 mL H ₂ O, 2mix 1 min, RT, 10 min; 3 extract with 8 mL ACN/FA 75:5; 4 Shake 15 min; 5 add 4 g anhydrous MgSO ₄ , 1 g NaCl, 1 g Na ₂ HCit 1.5 H ₂ O, Na ₃ Cit 2 H ₂ O, 6 Mix 1 min; 7 Shake 15 min; 7 Centrifuge to 2140 g, 2 min; 8 Filter; 9 Take 500 µL, dry; 10 Redissolved in 250 µL MeOH/H ₂ O 20:80	Raptor Fluoro Phenyl 50 x 2.1 n with Raptor Biphenyl 50 x 2.1 m Inj vol 10 μL, H ₂ O, 0.3% FA, B mM AmFc 20% B for 0.6 min, 20-40 % B i 8 min, keep 1 min, initial con	n, 2.7µm at 30 °C) MeOH, both with 5 n 0.4 min, 40-90% in	QQQ CaV 4.5 kV; ST 500 °C; CUR 40 psi; ISG 1 60 psi; ISG 2 65 psi
2020)				Flow: 0.4 mL/min	Time: t_{an} =10 min, t_{Tot} =13.5 min	LOQ: depen ding on the matrix , FBs 4-15 µg/kg
(A ure lie n De	B_1, B_2 and other myc	5	QuEChERS 1 Add 10 mL H ₂ O + 10 mL 0.5% AcOH in ACN; 2 Shake at 300 rpm, 5 min; 3 Add 5 g MgSO ₄ /NaCl 4:1, 4 Shake; 5 Centrifuge to 4000 g, 15 min, RT; 6 Take 5 mL; 7 Shake at 200 rpm, 5 min; 8 Centrifuge to 4000 g, 1 min; 9 Dry 1 mL at 40 °C (N ₂); 10 Add 75 μL MeOH; 11 Sonicate; 12 Add 75 μL H ₂ O, mix; 13 Centrifuge to 8500 g, 10 min, RT; 14 Dilute 60 μL with 140 μL H ₂ O; 15 Centrifugate	Zorbax Bonus-RP, 150 x 2 A) H ₂ O 0.15% FA, 10 mM Am FA 15% B 0.5 min, 15-100% B 8.5 15% B in 1 min, initial cond	Fo, B) MeOH 0.05% min, keep for 6 min,	QTra p SRM ST 550

sm	otoxi		to 8500 g, 10 min, RT			°C;
arc hel ier et al. 20 10)	ns Cere als					NG 50 psi; CUR 40 psi; TG 30 psi; CoG 1.2 x 10 ⁻⁴ psi
				Flow: 0.25 mL/min	Time: t_{an} =15 min, t_{Tot} =25.5 min	LOQ: 50 µg/kg all FBs
	B ₁ , B ₂ an d ot he r to xi ns			Ultra-Aqueous C ₁₈ , 100 x 2.1 mm, 3 μm, at 40 °C Inj vol 10 μL A) H ₂ O, B) MeOH, both with 0.1% FA+ 10 mM AmFo 10 % B for 1 min, 10-100% B in 6 min, keep for 3 min, initial conditions for 5 min		QTra p Condi tions NR
(Liao et al. 2013)	F1	1	1Add 5 mL H ₂ O/ACN 15:85; 2shake to 1550 rpm, 30 min; 3centrifugate to 4500 rpm, 5 min; 4take 500 μL; 5add 20 μL of ¹³ C-34 FB ₁ (25 μg/mL) + 480 μL 20 mM FA; 6vortex 15 s; 7filter	Flow: 0.5 mL/min	Time: t _{an} =10 min, t _{Tot} =15 min	LOD: FBs 2.2- 2.9 µg/kg, LOQ: FBs 7.3- 9.6 µg/kg, depen ding on the matrix
(Bart ók et al.		1	18 mL MeOH/H ₂ O 75:25; 2homogenize 9,500 rpm, 4 min: 3centrifuge to 10,000 rpm, 10 min, 4filter	YMC-Pack J'sphere ODS H80, 40 °C Inj vol 1 µI		TOF, full scan

2010) er s of B ₁ Ri ce		R	A) H ₂ O, B) ACN, both 24-40% B for 79 min, 40-100 ⁰ for 10 mir	% B for 15 min, keep	MS CaV 3.5 kV; Fragm entor 170 V; skim mer 70 V; DGT 350 °C; DGF 10 mL/m in; NG 20 psi; full scan 100- 1700; acquis ition rate 250 ms/sp ectru m
				Flow: 0.20 mL/min	Time: t_{an} =79 min, t_{Tot} =104 min	LOD/ LOQ: NR
(O ue sla ti et al. 20 12)	B ₁ , B ₂ Cere als, deriv ed prod ucts	5	1 Add 10 mL ACN/H ₂ O 80:20; 2vortex 2 min, shake 60 rpm x 10 min; 3centrifuge to 5000 rpm, 5 min; 4filter 2 mL (0.20 μm, Millipore)	Inj vol 5 μL, A) H ₂ O with 5 m 25-75% B in 3 min, 75-100% B	BEH C ₁₈ , 100x2.1 mm, 1.7 μm at 30°C I ₂ O with 5 mM AmFo, B) MeOH , 75-100% B in 2 min, keep for 1.5 1 min; initial conditions for 1 min	

				SS		350 °C; CGF 50 L/h; DGF 650 L/h
				Flow: 0.35 mL/min	Time: t_{an} =6.5 min, t_{Tot} =8.5 min	$\begin{array}{c} \text{LOD:} \\ B_1 \\ \text{and} \\ B_2 1 \\ \mu g/kg \\ \text{LOQ:} \\ B_1 \\ \text{and} \\ B_2 5 \\ \mu g/kg \end{array}$
(R au sc h, Br oc km ey er, an d Sc hw erd tle	B ₁ , B ₂ , B ₃ , HB ₁ , HB ₂ , HB ₃	2. 5	 Add ACN:H₂O:FA 79.20:1; shake for 15 min at RT; 3 Add 20 μL of Deuterated internal standard; 4 rotary agitation for 30 min; 5 centrifuge at 1902 g, 6 take an aliquot of supernatant, 7 filter 	First dimension: YMC-Pack Diol-NP C ₁₈ 100 × 2. mm, 5 µm at 40 °C. Vol. inj: 10 µL of sample A) H ₂ O, B) ACN:H ₂ O 90:10 Both (0.1% FA, 10 ml AmFo) 100% B in 2.5 min, 100-90% B in 0.5 min, 90-20 % B in 0.8 min, hold for 3.8 min, 20-100% B in 0.20 min. initial condition for 17.20 min. Second dimension: 2 columns connected in series Raptor FluoroPhenyl, 50 × 2.1 mm, 2.7 µm and Raptor Biphenyl 50 × 2.1 mm, 2.7 µm, 5% B for 1.2 min, 5-0% B in 0.10 min, hold for 7.1 min, 0-5% B in 0.05 min, 5-50% B in 1.1 min, 50- 70% B in 4.4 min, 70-85% B in 2.5 min, 85-100% in 3 min, hold for 2 min, 100-5% B in 0.10 min, initial condition for 4 min		QQQ CaV: 4.5 kV; CUR: 40 psi; ST: 500 °C;
20 21)				Flow 0.2 mL/ min, 0.3 ml/min	Time: tan= 7.6 min tTol= 25 min Time: tan= 15.50 min tTol= 25 min	LOQ: (B ₁₋₃ : 10, HB ₁₋₃ : 100) µg/Kg
			Other samples			
(ŠkrbB1 ić,I Add 40 mL ACN/H2O/AcOH 79:20:1; 2 Shake 1h; 3 Filter; 4 Take 20 mL; 5 Add 20 mL hexane; 6Hypersil GOLD C18, 50 x 2.1 mm, 1.9 µmŽivanB20Mix 2 min; 7 Centrifuge to 5000 rpm, 5 min; 8 Eliminate hexane phase. 9 Filter aqueous phaseFilter aqueous phaseČev,anS M B for 0.5 min, 5-95 % B in 2.5 min, k		OH, both with 1% AmAc	QQQ CaV 3.4 kV;			

and Godu la 2014	u ot he			95-5% B in 1.2 min, initial conditions for 1.8		ST 350 °C; SG 40 arbitra ry units; aux gas 10 arbitra ry units; CaT 270 °C
	ex tr ac ts of nu ts			Flow: 0.50 mL/min	t _{an} =6 min, t _{Tot} =8 min	LOD: 0.24 B ₁ , 0.05 B ₂ µg/kg LOQ: 0.8 B ₁ , 0.17 B ₂ µg/kg
(Y iba dat iha n, Jin ap, an d M ah yu din 20 14)	toxin s	B ₂ nd her xin		Symmetry C ₁₈ , 150 x 2.0 mm, 3μm, 30 °C Inj vol. 25 μL, A) H ₂ O, 0.2% FA, B) MeOH 10% B for 8 min, 10-90 % B in 2 min, keep 7 min, from 90-10% B in 3 min, initial conditions for 5 min		QQQ CaV 3 kV; ST 120 °C; DGT 350 °C
	Palm kern el cake	5	1 Add 20 ml H ₂ O/ACN/FA 20:79:1; 2 Shake 60 min; 3 Centrifuge supernatant to 3000 rpm, 10 min; 4 Dilute 1:4 with water; 5Filter	Flow: 0.20 mL/min	Time: t _{an} =17 min, t _{Tot} = 25 min	LOD both: Std 5.6 µg/kg LOQ both: Std 18 µg/kg LOD both: Sampl es

	P			5		17.5 μg/kg LOQ both: sampl es 58 μg/kg
(Q ian et al. 20 18)	B ₁ , B ₂ and other toxin s		QuEChERS	ACQUITY UPLC HSS T3, 100 x 2.1 mm, 1.8 μm at 40°C Inj vol. 5 μL A) H ₂ O, 0.1% FA, 1 mM AmAc; B) MeOH 0-10% B in 1 min, 10-20% B in 2 min, 20-99% B in 8 min, keep 2.5 min; 99-10% B in 0.1 min; initial		QQQ CaV 5.5 kV; ST 550°C ; Auxili
	Feed	2	1 Add 1.5 g NaCl + 10 mL 3% AcOH in ACN/H ₂ O 80:20; 2Vortex 1 min, 3Ultrasound 20 min; 4Add 2 g anh MgSO ₄ ; 5Vortex 1 min; 6Centrifuge to 8000 rpm, 5 min; 7Dry (N ₂ , 40 °C); 8Dissolve in MeOH:H ₂ O 1:1; 9Filter	Flow: 0.3 mL/min	Time: t_{an} =13.5 min, t_{Tot} = 18.5 min	ary gas 40 psi LOQ: 0.4 µg/kg for both
(Span jer, Rens en, and Schol ten 2008)	r to xi ns Pe an	2 5		Alltima C ₁₈ , 150 x 3.2 mm Inj vol 20 μL, A) H ₂ O, B) ACN 10-70% B in 12 min (curve 1), k in 1.5 min (curve 6), keep 2.5 min (curve 1), initial cond	I, both with 0.1% FA eep 4 min, 70-90 % B min, 90-10 % B in 1	B ₁ y B ₂ QQQ CaV 2.5 kV; CoV 75 V; DGT 450°C ; CGF 100 L/h (N ₂); DGF 600 L/h
	st ac hi o, w he at,		S	Flow: 0.3 mL/min	Time: t _{an} =20 min, t _{Tot} =25 min	LOQ: depen ding on the matrix , B ₁ 5- 100

	m ai ze , co m fl ak es , ra isi ns , fi gs			5 0 0		μg/kg, B ₂ 1- 100 μg/kg
(A uré lie n De sm arc hel ier et al. 20 14	B ₁ , B ₂ and other toxin s Cere als, coco a, oil, spice s, infan t form	2 5	 Add 50 mL H₂O, 2 Homogenize 1 min 10000 rpm, 3Take 5 g of sample (peanut, green cofee, cocoa, paprika) or 2 g (infant formula, sunflower oil), 4Add 100 μL of ¹³C-FB standard (FB₁ and FB₂ each 10 μg/mL), 5Add 10 mL H₂O and 10 mL ACN, 0.5% AcOH, 6 Add 5 g MgSO₄:NaCl 4:1 Centrifuge 4000g, 15 min, 7Defat 5 mL ACN phase with 5 ml hexane. 8 Take 1 mL of ACN phase, dry, 9Reconstitute in 150 μL H₂O/MeOH 1:1, 10Centrifuge 8500 g, 10 min, 11Take 60 μL, add 140 μL H₂O, 12Centrifuge 8500 g, 10 min 	Zorbax Bonus-RP C ₁₈ , 150 x 2.1 mm, 3.5 μm at 50 °C Inj vol 20 μL A) H ₂ O, 0.15% FA, 10 mM AmFo, B) MeOH, 0.05% FA 15% B for 0.5 min, 15-100 % B in 6 min, keep for 4.5 min, 100-15% B in 0.5 min, initial conditions for 7.5 min		QTra p, QQQ ST 550 °C; CUR 40 psi, Nebul izer 50 psi; Turbo gas 30 psi
)	ula, coffe e, nuts			Flow: 0.35 mL/min	Time: $t_{an}=11$ min, $t_{Tot}=19$ min	LOD/ LOQ: NR
(S har et al. 20 20)	B ₁ , B ₂ and other toxin s Feed , its ingre dient s	5	1 Add ACN/H ₂ O/FA 79:20:1; 2 Shake for 90 min to 180 rotations/s; 3Centrifuge to 4000 rpm, 2 min, 4 Filter	Acquity C_{18} , 100 x 2.1 mm Inj vol 20 µl A) H ₂ O, 1% FA, B) MeOH/H ₂ with 10 mM Ar 0% B for 2 min, 0-50% B in 0.5 3.5 min, keep 1 min, initial conc wash for 5 m	L O/FA, 97:2:1, both nFo. 5 min, 50-100% B in ditions in 1 min, seal	sQ CaV 2.79 kV; ST 150 °C; DGT 350 °C; CGF 50

	B ₁ , B ₂			Flow: 0.5 mL/min	Time: <i>t</i> an=7 min, <i>t</i> Tot= 8 min	CGF 600 L7h LOD B ₁ : 0.07 µg/kg, LOQ B ₁ : 0.22 µg/kg LOD B ₂ : 0.03 µg/kg, LOQ B ₂ : 0.03 µg/kg, LOQ B ₂ : 0.08 µg/kg, LOQ
(Fr eni ch et al. 20 09	B ₂ Maiz e, waln ut, brea kfast cere	5 1 Add 10 mL ACN/H ₂ O 80:20 (for biscuit add 20 mL); 2 Vortex 2 min; 3. Centrifuge to 4500g, 5 min; 5 Take and filter 2 m	1 Add 10 mL ACN/H ₂ O 80:20 (for biscuit add 20 mL); 2 Vortex 2 min; 3 Shake to 60 rpm, 10 min; 4 Centrifuge to 4500g, 5 min; 5 Take and filter 2 mL	Acquity C ₁₈ , 100 x 2.1 mm Inj vol 5 μ L, A) H ₂ O with Aml 25-75% B in 3 min, 75-100% B min, 100- 25% B in 1 min; initia	Fo 5 mM, B) MeOH in 2 min, keep for 1.5	Cav 3.5 kV; EV 3 V; ST 120°C ; DGT 350°C ; CGF 50 L/h; DGF 650 L/h (N ₂ for both)
)	al, bisc uit			Flow: 0.35 mL/min	Time: t_{an} =6.5 min, t_{Tot} =8.5 min	$\begin{array}{c} LOD\\ maize\\ : B_1\\ 0.1\\ \mu g/kg,\\ B_2 0.2\\ \mu g/kg,\\ LOQ\\ maize\\ : B_1\\ 0.5\\ \mu g/kg, \end{array}$

						$\begin{array}{c} B_2 \ 0.6 \\ \mu g/kg; \\ LOD \\ breakf \\ ast \\ cereal \\ \vdots \ B_1 \\ 2.1 \\ \mu g/kg, \\ B_2 \ 0.7 \\ \mu g/kg \\ LOQ \\ breakf \\ ast \\ cereal \\ \vdots \ B_1 \\ 6.2 \\ \mu g/kg, \\ B_2 \ 2.5 \\ \mu g/kg \end{array}$
			Beverages			
(R ub ert al. 20 11)	B ₁ , B ₂ , B ₃ and other toxin s	1 0 m L	1 Sonicate 25 min, 2Condition SPE Oasis HLB cartridges with 5 mL ACN/MeOH 1:1; 3 5 mL H ₂ O; 4 10 mL sample into cartridge; 5Wash with 5 mL H ₂ O; 6 Dry 30 min; 7 Eluate with 4mL ACN:MeOH 1:1; 8 Dry (N ₂ , 35 °C), 9 Reconstitute in 1 mL (ACN/MeOH 1:1); 10Filter	Gemini C ₁₈ , 150 x 2.0 mm. Inj vol 10 µl A) H₂O, 0.1% FA, B) MeOH, bc 5-95% B in 10 min, 95-80% conditions 5 n	L oth with 5 mM AmFo B in 5 min, initial	QQQ Orbitr ap XL CaV 30 V; SV 4 kV; Sourc e Temp 275 °C; Capill ary gas sheat 35 units; auxili ary gas 30 arbitra ry units
				Flow: 0.2 mL/min	Time: $t_{an}=10$ min, $t_{Tot}=20$ min	LOD: 30-35 µg/L,

				65		LOQ: 90- 105 µg/L all Fbs depen ding of the beer type
(H ua ng et al. 20 18)	B ₁ , B ₂ and other toxin s	2	QuEChERS 1Add 100 µL of D-atrazine (60 µg/L), 15 mL acetate buffer pH 3.0, 10 mL 5% FA in ACN; 2 Shake; 3 Extract with ultrasonic (53 KHz, 5 min, 20°C); 4 Add 4 g MgSO ₄ + 1 g NaCl + 0.5 g Na ₂ HCit·1.5H ₂ O, 1 g Na ₃ Cit·2H ₂ O; 5 Shake to 1500 strokes/min, 5 min; 6 Ice bath 10 min, 7Centrifuge to 18514 g, 10 min; 8 Take 6.0 mL; 9 Transfer supernatant into 15 mL centrifugation tube containing 900 mg MgSO ₄ , 600 mg C ₁₈ , 150 mg PSA, 150 mg Si; 10 Shake 5 min, 11 Centrifuge 10 min; 12 Take 2 mL, reduce volume <0.5 mL with N ₂ ; 13 Complete to 1 mL with H ₂ O/MeOH 80:20; 14filter	Poroshell EC-C ₁₈ , 150 x 3 mi A) H ₂ O, B) MeOH, 0.2% FA Inj vol 5 μI 20% B for 2 min, 20-50% B in 2 min, keep 1 min, 100-20% I conditions for 2	A and 2 mM AmF min, 50-100% B in 7 3 in 1 min, initial	QQQ CaV 5.5 kV; DP 150 eV; EP100 eV; CUR 30 psi; GS1: 50 psi, turbo gas (gas 2) 50 psi, GT: 450°C
				Flow: 0.45 mL/min	Time: t_{an} =12 min, t_{Tot} = 15 min	LOD: B ₁ , B ₂ , 0.05 µg/kg LOQ: B ₁ . B ₂ , 0.125 µg/kg
(T am ura et al.	B_1, B_2, B_3 and other	5 m L	1Add 25 mL AmAc 10 mM, mix, 2wash in Oasis HLB SPE Cartridge conditioned with 5 mL AmAc 10 mM, 3elute with 5 mL AmAc 10 mM/ACN 1:1, 4elute 5 mL ACN, mix, dry N2 40°C, 5dissolve in 1mL H ₂ O, 660 μL FA + 5 mL ACN, mix, 7apply to multistep #229 Ochra cartridge. 8Dry 4 mL of eluate with N ₂ 40°C, 9dissolve in 500 μL AmAc 10 mM/ACN 85:15, 10filter	Acquity UPLC BEH C ₁₈ , 100 s 40°C A) H ₂ O; B) MeOH, with 2 AmA Inj vol 5 μI	% AcOH, 0.1 mM c	QQQ CaV 3 kV; ST 120°C

12)	toxin s			5		450 °C; CGF 50 L/h; DGF 800 L/h
	Win e			Flow: 0.3 mL/min	Time: t_{an} =5 min, t_{Tot} = 7 min	LOD: 0.30 µg/L, LOQ: 1 µg/L all Fbs
	B ₁ , B ₂ and other toxin s					QQQ CaV 4 kV; DGF 18 L/min
(M iró - Ab ell a et al. 20 17)	Plant - base d beve rage	1 0 m L	 Add 10 mL 1% FA in ACN in a 50 mL centrifuge tube, 2 Shake 3 min; 3 Add 4 g MgSO₄ + 1 g NaCl; 4 Shake vigorously 3 min; 5 Centrifuge to 10000 rpm, 5 min, 20°C, 6dilute 1:1 with phase A 7filter 	Cortecs UHPLC C ₁₈ , 100 x 2.1 Inj vol 5 A) H ₂ O, B) MeOH, both with AmAc 10-50% B in 4.5 min, 50-95% min	μL 0.1% AcOH, 5 mM	; DGT 160°C ; nebuli zer 35 psi; nozzle voltag e 0.5 kV; Frag Vol 380 V
	S		Flow: 0.45 mL/min	Time: t_{an} = 14.5 min, t_{Tot} = NR	LOD: 0.80; LOQ: 2.68 µg/kg all Fbs	
(B. Zh an g et	B ₁ and other toxin s	5	1 Add 5 mL distilled H ₂ O, 10 mL 1% AcOH in ACN; 2 Shake to 3000 rpm; 3 Add 1 g NaCl + 4 g MgSO ₄ , 4 Centrifuge to 13000 rpm, 5 min, 10 °C; 5 Transfer into 10 mL polypropylene tube containing 450 mg MgSO ₄ ; 6 Shake 30 s; 7Centrifuge to 5000 rpm, 5 min, 10 °C	ZORBAX RRHD Eclipse Plus C_{18} , 50 x 2.1 mm, 1.8 µm at 30°C, Inj vol 2 µL A) H ₂ O, B) ACN, both with 0.1% FA 10-42% B in 2.4 min, 42-51% B in 3.6 min, 51-95% B in 0.2 min, 95-10% B for 0.8 min, initial conditions		QQQ CaV 4 kV; DG tempe

al. 20 18)	Grap es, wine s			for 5 min Flow: 0.3 mL/min	Time: $t_{an}=6.2 \text{ min},$ $t_{tot}=12 \text{ min}$	rature 350 °C; DG flow 10 L/min ; Nebul izer 40 psi LOD: 1 μg/L, LOQ:
(Pi	B_1, B_2, B_3 and other toxin s			Let T _{rot} = 12 mm Acquity UPLC BEH C ₁₈ , 100 x 2.1 mm, 1.7 μm, 50 °C Inj vol 5 μL		2001 3 μg/L QQQ CaV 2 kV; ST 120 °C; DGT 400
zz utt i et al. 20 14)	Win es	5	1 Add 5 mL H ₂ O, 10 mL 1% AcOH in ACN, 25 μg/mL of: FB ₁ (ACN/H ₂ O 1:2), FB ₂ (CAN/H ₂ O 1:3), and FB ₃ (ACN); 2Mix to 300 rpm, 1 min; 3 Add 3 g anh. MgSO ₄ ; 4 Shake 1 min; 5 Centrifuge 13000 rpm, 5 min, 6Take 3 mL of superior phase; 6 Mix with 450 mg anh. MgSO ₄ ; 7 Mix 10 s, centrifuge 4000 rpm, 4 min, 10 °C; 8 Filter and dilute 1:1 with MeOH	A) H ₂ O, B) ACN, both v 10-70% B in 10 min, 90 %	vith 0.1% FA 6 B for 2 min, initial s for 1 min	400 °C; DGF 100 L/h; CGF 700 L/h LOQ: 50
				Flow: 0.4 mL/min	Time: $t_{an}=12$ min, $t_{Tot}=13$ min	μg/kg all Fbs
(P ére z- Or teg a et al. 20 12)	B ₁ and other toxin s Win e	4 m L	Oasis HLB, Bond Elut Plexa 1 SPE cartridges preconditioned with 4 mL MeOH, 2 4 mL H ₂ O at 2 mL/min; 3 Add sample into cartridge; 4 Elute with MeOH/H ₂ O 5:95; 5 Dry in vacuum 1 min; 6 Elute twice/4 mL MeOH, 1 mL/min; 6 Evaporate (N ₂ , 37°C); 7Reconstitute (1 mL MeOH:H ₂ O 2:8); 8 Filter	Zorbax Eclipse XDB-C ₁₈ , 50 x 4 NR Inj vol 20 μ A) H ₂ O, 0.1% FA; 10 % B for 2 min, 10-50% B in 10 min, keep 3	L B) ACN 3 min, 50-100% B in	TOF CaV 4kV; NGP 40 psi; DGF 9 L/min ; DGT 325 °C;

				Flow: 0.5 mL/min	Time: $t_{an} = 18 \text{ min},$ $t_{Tot} = \text{NR}$	Frag Vol 190 V; range 50 - 1000 LOD: 0.8 µg/L, LOQ: 2.68 µg/L
			Samples of animal origin			
(C ao et al. 20 18)	B ₁ , B ₂ Urin e, plas ma	2 0 0 μ L ur in e 2 0 0 μ	 Add 50 μL β-glucuronidase + 20 μL SI (¹³C₃₄-FB₁ 1 mg/mL); 2incubate 37 °C overnight; 3centrifuge to 10000 rpm, 5 min; 4take supernatant, add 730 μL H₂O/ACN 90:10; 5filter Add 50 μL β-glucuronidase + 20 μL SI (¹³C₃₄-FB₁ 1 mg/mL); 2incubate 27°C overnight; 3add 1mL ACN:AcOH 99:1; 4vortex 30 s; 5centrifuge to 5000 rpm, 10 min; 6dry at 45°C; 7reconstitute in 200 μL of H₂O:ACN 9:1; 8mix 30 s; 9filter 	Kinetex C ₁₈ , 100 x 2.1 mr Inj vol 10 μ A) H ₂ O, 0.2 mmol 25% B for 1 min, 25-70% B ir 0.5 min, initial conditio	L L AcOH; B) MeOH 2 min, 70-25% B in	QQQ, TISP CUR 20 psi; CoGG (CAD) 8 psi; GS1 20 psi; GS2 15 psi; GT 600°C ; EP 10.0; CP 12.0
		L pl as m a		Flow: 0.2 mL/min	Time: t_{an} =3 min, t_{Tot} = 5 min	LOD B ₁ : urine 0.12 µg/L, LOQ B ₁ : urine 0.45 µg/L LOD B ₁ :

plasm a 0.19 μg/L, LOQ B_1 : plasm a 0.39 μg/L 000 B_1 CaV 4 and kV, other toxin ST 300 s °C; Hypersil Gold C₁₈, 50 x 2.1 mm, 1.9 µm at 45 °C Aux Inj vol 2.5-10 µL, A) H₂O with 0.1% AcOH, B) (D gas 18 MeOH evr au; 2 5 35 % B for 1.5 min, 90 % B in 0.5 min, keep 1.5 min, ees ISGP 90-35 % B in 0.2 min, initial conditions 2.3 min 1.- Add 12.5 μL ¹³C-34 FB₁ (25 μg/mL in ACN) + 750 μL ACN (deproteinization); 2.-vortex 15 s; 3.e 4 au; 0 centrifuge to 8517 g, 10 min, 4°C; 4.-evaporate supernatant (N2, 45 °C); 5.-reconstitute with 200 µL et SGP al. H₂O/MeOH 85:15; 6.-vortex 15 s, 7.-filter μ 23 au; Pig 20 L VT plas 12 300 ma °C; LOD: 0.8 Time: t_{an} =3.5 min, μg/L, Flow:0.30 mL/min LOQ: $t_{\text{Tot}} = 6 \min$ 1 μg/L (A QQQ B_1 , ST rro B_2 500 yo °C: М Zorbax Eclipse C₁₈, 50 x 2.1 mm, 1.8 µm at 35 °C CUR Inj vol 5 µL, A) H₂O, B) MeOH, both with 0.3% FA, 30 an 5 mM AmFo psi; za Milk **OuEChERS** 5-50% B in 1 min, 50-72 % B for 2 min, 72-80 % B ISV 5 nar thistl 1.- Add 8 mL of 30 mM NaH₂PO₄ (pH 7.1); 2.-vortex 10 s; 3.-add 5 mL ACN with 5% FA; 4.- shake 2 min; for 2 min, 80-90 %B for 2 min, 90-5% B in 0.2 min kV; es, 2 e 5.-sdd 4 g MgSO₄ + 1 g NaCl + 1 g NaCit + 0.5 g Na₂HCit 1.5 H₂O; 6.- shake 1 min; 7.-centrifuge to 4500 Ga gas 1 Silyb rpm, 5min); 8.-take 1 mL; 9.- dry; 10.-reconstitute with 1 mL MeOH/H₂O 1:1; 11.-filter and rcí um gas 2 amari 50 psi Ca anu mp LOD: т añ B₁ 3.9 a, Flow: 0.4 mL/min Time: $t_{an}=7.2 \text{ min}$ μg/kg, an 13.7 d

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ug/kg

Gá LOQ: mi B_1 13.5 Z-Gr μg/kg, B2 aci 45.7 а 20 µg/kg 13) QQQ B₁, CaV B₂, 2.5 B_3 kV; CoG: CORTEX C18 10 x 4.6 mm, 5 µm at 40 °C 0.15 Vol. Inj NR mL/ A) H₂O B) MeOH both with 0.2 % -FA min, (S. 10-90%B in 6 min; hold for 2 min; initial condition DGT Zh for 2 min 500 Broil an °C: 1.- Add 20 mL of ACN:H₂O; 2.- shake for 30 min; 3.- ultrasonic for 30 min; 4.- take 50 µL; 5.- centrifuge at g er DGF: et Chic 5 8000 rpm for 15 min; 6.- add 950 μ L of H₂O and vortex; 7.- take 50 μ L; 8.- add 10 μ L of IS ¹³C-FBs; 9.-800 dilute with 850 μ L of in MeOH:H₂O 1:9 (0.2 % -FA) al. ken L/h; 20 Feed LOD: 22 and 50) Excr µg/Kg eta all Time: t_{an}= 8 min Fbs Flow 0.4 mL/min $t_{Tot} = 10 \min$ LOQ 160 µg/Kg all Fbs Qtrap B₁, CaV: B_2 5.5 kV; Shimadzu C₁₈ 100 × 2.1mm, 1.8 μ m at 40 °C (W CoG: Vol. Inj. 3 µL of sample 35 eiy A) H₂O (1% FA), B) ACN 1.- Add IS (13C34-FB1 ing psi; 117 ($^{13}C_{34}$ -FB₁), $^{13}C_{34}$ -fumonisin B₂($^{13}C_{34}$ -FB₂) mixed internal standard (25 µg/mL); 2.- add 5 mL of 5% B for 1 min; 5 -90 %B in 3.5 min; hold for 2.5 CUR: et ACN:H₂O (2% FA); 3.- vortex for 10 min; 3.- Centrifuge at 3900 rpm for 3 min; 4.- evaporate to dryness at min; initial condition in 0.1 min; hold for 1.9 min 1 al. 35 40 °C under N2; 5.- redissolved in 5 mL of H2O; 6.- Add 6 mg of DSPME MIL-101 (Cr); 7.-ultrasonic for 10 Milk 20 psi; min; 8.- centrifuge at 1200 rpm for 5 min; 9.- filter GS2: 22 45 psi) LOD: Time: $t_{an} = 8 \min$ 1.5 Flow 0.4 mL/min $\mu g/Kg$ t_{Tot}= 10 min all Fbs

				1 6		LOQ 5 µg/Kg all Fbs
(Fl ore s- Fl ore s an d Go nz ále z-	B ₁ B ₂ , B ₃ Milk	2, 3 3 1 m I	I Add 4 mL 2 % FA in ACN, 2shake 15 min; 3centrifuge 5000 rpm, 10 min, 4take 4 mL supernatant,	Ascentis Express C ₁₈ , 150 x 2.1 mm, 2.7 μm, 45°C Inj vol 20 μL, A) H ₂ O, B) MeOH/H ₂ O 95:5, both with 0.1% FA and 5 mM AmFo 5-28% B in 5 min, 28-45 in 5.5 min, 45-60% B in 0.5 min, 60-90% B in 5 min, keep for 1 min, initial conditions for 13 min		QQQ CaV 4 kV; DGT (high- purity N ₂) 350°C ; DGF 9 L/min ; 275.8 Pa, dry gas 40 psi
z- Pe ña s 20 18)				Flow: 0.4 mL/min	Time: 16 min	LOD/ LOQ: FB ₁ 10 µg/L, FB ₂ 2.5 µg/L, FB ₃ 0.625 µg/L
	B ₁ and other toxin s					QQQ CaV 3.2 kV; DGF
(S on g et al. 20 13)	5 m Pig, L hum an urine		m aqueous phase, add 5 mL ACN/FA 99:1; 4repeat extraction; 5dry (N_2 , 60°C); 6reconstitute with 500 μ L	Symmetry C ₁₈ , 150 x 2.1 mm, 5 μm at RT Inj vol 20 μL A) H ₂ O, B) MeOH, both with 0.3% FA, 5 mM AmFo 5% B for 1 min, 5-25% B in 4 min, 25-60%B in 2 min, 60-80% B in 8 min, 80-100 B in 1 min, keep 6 min, 100-5 % B in 3 min		800 L/h; CGF 20 L/h; DGT 350 °C; ST 120 °C
				Flow: 0.25 mL/min	Time: $t_{an}=22 \text{ min}$, $t_{Tot}=25 \text{ min}$	LOD: 0.05

				39		ng/m L, LOQ: 0.17 ng/m L
(K. Zh an g et al. 20 13)	B ₁ , B ₂ , B ₃ and other toxin s	_		Phenomenex Kinetex XB-C ₁₈ , 140° C Inj vol 5 μ I A) H ₂ O, B) MeOH, both with		QQQ -IT CaV 5.5 kV; CUR 30 psi;
	Milk base d infan t	0. 5	 Add 25 μL IS (¹³C₃₄ FB₁, ¹³C₃₄ FB₂, ¹³C₃₄ FB₃ 500 ng/mL); 2 Vortex 30 s; 3 Add 5 mL ACN/H₂O 1:1; Shake 10 min at 30-35 pulsations/min; 5 Take an aliquot of 2 mL; 6 Filter 2 mL; 7Centrifuge to 4500 rpm, 30 min 	AmFo 5-40% B lineal in 2 min, 40-100 min, keep 2.5 min, 100-5% E conditions for 3	3 in 0.5 min, initial	ST 450 $^{\circ}C;$ gas 1 and gas 2 60 psi
	food s			Flow: 0.3 mL/min	Time: t_{an} =11.5 min, t_{Tot} = 15 min	LOQ B ₁ : 2 µg/kg all fbs
(Abia xi Abia)		1 m	1 Centrifuge to 5600 g, 3 min; 2take 100 µL 3add 900 µL H ₂ O/ACN 9:1	Gemini 150 x 4.6 m Inj vol 5 μL, A) H ₂ O, B) ACN, b 5 % B for 2 min, 5-30 % B in 8 min, keep 1 min, initial condi	ooth with 0.1% AcOH min, 30-96 % B in 4	QTra p ST 650 °C, CUR 30 psi, SG 80 psi, DG
(Aola et al. 2013)		r to xi ns 1 m L U ri		Flow: 0.6 mL/min	Time: t_{an} =15 min, t_{Tot} = 17.25 min	$\begin{array}{c} 80 \text{ psi} \\ \text{LOD:} \\ B_1 \\ \text{and} \\ B_2 \ 0.5 \\ \mu g/L, \\ \text{LOQ:} \\ B_1 \\ \text{and} \\ B_2 \ 1.7 \\ \mu g/L \end{array}$

(N ual ka w et al. 20 20)	B ₁ , B ₂ and other toxin s Swin e, Poul try, Dair y	1	QuEChERS 1Add 10 mL H ₂ O 1% FA, 2soak 30 min; 4add 10 mL ACN: 5shake to 240 rpm, 30 min; 6add 1 g NaCl + 4 g MgSO ₄ ; 7shake 30 s: 8 centrifuge to 10000 rpm, 5 min; 9take 2 mL; 10add 0.1 g silica C ₁₈ + 0.3 g MgSO ₄ ; 11mix; 12 centrifugate 1 min; 13dry at 40 °C, 14reconstitute in 960 μL MeOH 20% + 40 μL (250 ng/mL ¹³ C-34 FB ₁ +50 ng/mL ¹³ C-34 FB ₂); 15filter	Accucore C_{18} , 100 x 2.1 mi Inj vol 3 µI A) deionized H ₂ O, 0.1% FA, 51 0-20% B in 4 min, 20-40% B in in 10.5 min, keep 2.5 min; initia	nM AmF; B) MeOH 5.5 min; 40-100% B	Qtrap Needl e voltag e 4.5 kV; CUR 30 psi; nebuli zer (Gas1), turbo gas (Gas2) 55 psi; turbo gas tempe rature
	Feed s			Flow: 0.4 mL/min	Time: t_{an} =22.5 min, t_{Tot} = 25.5 min	$\begin{array}{c} 500 \\ ^{\circ}C \\ LOD: \\ B_1 15 \\ \mu g/kg, \\ B_2 4.5 \\ \mu g/kg; \\ LOQ: \\ B_1 30 \\ \mu g/kg, \\ B_2 9 \\ ng/kg \end{array}$
(O ste res ch et al. 20 17)	B ₁ and other toxin s	1		Gravity SB C ₁₈ , 100 x 2.0 n	um, 3 μm at 45°C	QTra p CaV 5.5 kV;
	Bloo d or seru m	1 0 μ L	LLE 1Spott 4 times on filter paper; 2dry overnight at RT, 3Extract with 1 mL H ₂ O/acetone/ACN 30:35:35 in 2 mL safe-lock tubes; 4Sonicate 30 min; 5Take 800 μL; 7Dry at 50°C under reduced pressure; 8 Reconstitute with H ₂ O/ACN/AcOH 95:5:0.1; 9Centrifuge to 22000 g, 10 min	Inj vol 30 μ A) H ₂ O, 0.1% AcOH, B) A 3-15% B in 3 min, 15-55% B in min, 55-100% B in 2 min, k conditions 1.5	L ACN, 2% AcOH 1.5 min, keep for 1.5 eep 10 min, initial	ST 500 °C; DP 125 V; CUR 40 psi; GS1

	6		45 psi; GS2 50 psi					
	Flow: 0.75 (0-6), 0.85 (6.1- 10), 0.75 (10.1-11.5) mL/min	Time: t_{an} =10 min, t_{Tot} = 11.5 min	LOD: 0.521 ng/L LOQ: 2.5 ng/m					
 (ACN) Acetonitrile, (AcOH) Acetic acid, (AE) Appearance energy, (AmAc): ammonium acetate, (AmFo) Ammonium formate, (CaV) Capillary voltage, (CaT) Capillary temperature, (CGF) Cone gas flow, (CoG) Collision gas, (CUR) Curtain gas, (DG) Drying gas, (DGF) Desolvation gas flow, (DGT) Desolvation gas temperature, (EV) Extractor voltage, (FA) Formic acid, (Frag Vol) Fragmentor Voltage, (GF) Gas flow, (GT) Gas Temp, (LIT) linear ion tramp, (MeOH) Methanol, (MSPD) Matrix Solid Phase Dispertion, (NG) Nebulizer gas, (NR) Not reported, (PLE) Pressurize Liquid Extraction, (RT) Room temperature, (ST) Source temperature, (SV) Source voltage, (t_{an}) analysis time, (t_{rot}) total time including column conditioning. 								

			Table 2. LC-MS methods for FBs with clean up					
	F Bs	S a m pl e	Sample treatment	LC conditions	MS condi tions			
Ref	M at ri x	(g)	Extraction procedure	Column / Injection volume / Mobile Phase Flow / Analysis Time	Mass Condit ions / Limits			
	Maize and corn-based products							
(Re	B ₁ , B ₂ , B ₃		1 Add 200 μL of IS (2.5 μg/mL ¹³ C ₃₄ - FB ₁ , 1 μg/mL ¹³ C ₃₄ -FB ₂ , ¹³ C ₃₄ -FB ₃); 2 Add 10 mL ACN/H ₂ O 1:1; 3 Extract with ultrasonic 1h; 4 Centrifuge to 15 000 rpm, 6 min; 5 Adjust pH to 7-9 with NaOH; 6 Take an aliquot of 3 mL; 7Dilute with MeOH/H ₂ O (66.7:33.3)	BEH C ₁₈ , 100 x 2.1 mm, 1.7 μm, 35°C Inj vol 2 μL A) H ₂ O, 0.1% FA, B) ACN/MeOH 1:1 30-70% B in 2.3 min, 70% B for 1.7 min, 70-100% B in 0.2 min, keep for 0.6 min, 100-30% B in 0.2 min., re-equilibrate for 2 min	QQQ CaV 3.5 kV; CoV 45 V; ST: 120; CGF: 50L/h DGT 350°C; DGF 500 L/h			
n et al. 201 1)	Ma ize		a ⁵ 1 Load the dilute sample in MultiSep 211 FUM cartridge; 2Pass 8 mL of MeOH/H ₂ O (66.7:33.3); 3 Pass	Flow: 0.3 mL/min Time: t_{an} =4.8min, t_{Tot} =7 min	$\begin{array}{c} LM\\ LOD:\\ (B_1\\ 0.45,\\ FB_2\\ 0.50,\\ B_3\\ 0.10)\\ \mu g' kg\\ LOQ:\\ (B_1\\ 1.50,\\ FB_2\\ 1.65,\\ B_3\\ 0.40)\\ \mu g' kg \end{array}$			
(L. Silv	B_1, B_2 an	25	1 Add 40 mL MeOH/H ₂ O 80:20; 2 Centrifuge to 2500 g, 15 min 3 Extract the remaining solid with 30 mL MeOH/H ₂ O 80:20; 4 Filter	Luna C ₁₈ , 250 x 4.6 mm, 5 µm Inj vol 10 µL A) H ₂ O, B) MeOH both with 0.5% FA	QQQ CaV 4 kV,			

a et al. 200 9)	d oth er tox ins			65% B for 4 min, 65-95% B in 4 min, ke		GT 350°C; DGF 13 L/min;
	Co rn- bas ed pro du cts		Clean up: 1 Dilute 10 mL of filtrate with 40 mL of PBS; 2 Take 20 mL 3 Add to a FumoniTestTM immunoaffinity; 4 Wash with 10 mL PBS; 5 Eluted twice with 1.5 mL of MeOH; 6 Evaporate (N ₂ , 60 °C); 7 Reconstitute in 50 µL MeOH/H ₂ O (1:1)	Flow: 0.50 mL/min	Time: $t_{an} = t_{Tot} = 15 \text{ min}$	NG 30 psi; LOD: 40 µg/kg, LOQ:1 10 µg/kg all Fbs
(Ca	B_1, B_2 an d oth er tox ins		1 10 mL ACN/H ₂ O 75:25; 2 homogenize 15s; 3 Transfer on cartridge (6 mL) with 100 mg of C_{18} ; 4Wash the extract with 7 mL of ACN/H ₂ O 75:25, twice; 5 Collect 25 mL; 6 Take 5 mL; 7 Dilute with 500 mL of H ₂ O.	Alltima C18, 250 x 2.1 mm, 5 μ m, 45 °C Inj vol 20 μ LA) H2O, B) MeOH, both containing 25 mmol/L FA, adjusted to pH 3.8 with ammonia 60% B for 3 min, 60-90% B in 5 min, 100% for 10 minFlow: 0.2 mL/minTime: $t_{an} = t_{Tot} = 18$ minGemini C18, 150 x 2 mm, 5 μ m, 40°C Inj vol 20 μ L A) H2O (0.5% AcOH, 1 mM AmAc) B) MeOH (0.5% AcOH, 1 mM AmAc 20-40% B in 3 min, 40-63% B in 35 min, keep constant for 11 min, initial conditions for 10 min		QQQ CoV 5.5 kV; CUR 35; GS1 35;
vali ere et al. 200 7)	Ma ize	1	Clean up: 1 Load sample dilute on SPE-Carbograph-4 (500 mg); 2 Wash with 10 mL of H ₂ O; 3 Pass 0.3 mL of MeOH; 4 Elute with 1 mL MeOH and 8 mL of DCM:MeOH 8;2 (50 mM of FA); 5 Evaporate to 100 µL; 6 Add IS (FB ₁ , FB ₂ in MeOH/H ₂ O 1:1 (1 mg/mL); 7 Evaporate to 100 µL; 8 Dilute with 100 µL of LC mobile phase			GS2 40; GT 350 °C LOD/L OQ: 10 mg/kb for FB1 and 5ng/kg FB ₂
(Lat tanz io et al. 200 7)	$\begin{array}{c} B_1,\\ B_2\\ an\\ d\\ oth\\ er\\ tox\\ ins \end{array}$	10	 1 Add 50 mL de PBS; 2 Shake 60 min; 3 Centrifuged to 3000 g, 10 min; 4 filtrate 35 mL of PBS (extract A); 5 Add 35 mL of MeOH, to the remain solid, containing 15 mLPBS; 6 extract again 7 Shake 60 min; 7 Centrifuge to 3000 g, 10 min; 8 Dilute 10 mL of extract with 90 PBS (extract B); 9 Filter 			QTrap GT 350 °C; CUR 30 PSI; CoV: 4.5 kV:

	Ma ize		Clean up: 1Load 50 mL of extract B to the IAC; 2 Wash with 20 mL of PBS; 3 Add 5 mL of extract A; 4 Wash with 10 mL of water; 5 Eluate both extracts with 1.5 mL MeOH twice; 6 Dry at 50 °C; 7 Reconstitute with 200 µL MeOH/H ₂ O 4:6 (1 mM AmAc and 0.1% AcOH)	Flow: 0.200 mL/min	Time: t_{an} =49 min, t_{Tot} =59 min	GS1: 10 psi, GS2 30 psi. LOD: B ₁ 1.1 μg/kg, B ₂ 0.4 μg/kg
(Y. Wa ng et al. 201 3)	B ₁ an d oth er tox ins		 Add 50 mL of ACN/H₂O/AcOH (79:20:1); 2 Stir for 10 min; 3 Filter; 4- Evaporate 10 mL to dry; 5 Redissolve in 100 μL of MeOH; 6 Vortex 1 min; 7 Add 1.9 mL of H₂O 8 Vortex again for 1 min 	In A) H ₂ O, B) MeOH t 50% B for 5 min, constant for 10 min	5 75 x 3.0 mm, 2.2μm, 30°C j vol 20 μL both with 0.1% AcOH, 1 mM AmAc 50-10% B in 5 min, keep n, 10-50% B in 1 min, keep	QTrap GT 450°C; CUR 10 psi; GS1 50 psi; GS2
	Ma ize	10	Clean up: 1 Active the Oasis HLB SPE cartridges with 2 mL of MeOH; 2 Equilibrate with MeOH/H ₂ O (05:95); 3 Load sample; 4 Wash with 2 mL MeOH/H ₂ O (05:95); 5 Elute with 2 mL of MeOH; 6 Dry (N ₂ , 50°C); 7 Redissolve in 1 mL MeOH/H ₂ O (2:8)	cons Flow: 0.30 mL/min	tant for 4 min Time: $t_{an}=21 \text{ min}, t_{Tot}=25 \text{ min}$	50 psi; SV 5.5 kV LOD: 0.64 μg/kg, LOQ: 2.12 μg/kg
	D		Other cereals and seeds			
(Br yła, Ren ata, et al. 201	B ₁ , B ₂ , B ₃ Ce rea 1 pro du	25	 1Add 100 mL ACN/MeOH/H₂O (25:25:50); 2 Stir 30 min; 3 Centrifuge to 10730 g, 10 min; 4 Dilute the supernatant 1:1 with 10 mL deionized H₂O Clean up: 1 Transfer 8 mL of dilute extract to a FumoZon cartridge; 2 Preconditionate with 4 mL of MeOH and H₂O; 3 Wash with 6 mL ACN/H₂O (25:75); 4 Eluate with 4 mL of 2% FA in MeOH; 5 Evaporate to dry; 6 Redissolve in 1 mL of MeOH/H₂O/AcOH (1:8.9:0.1) 	 Kinetex PFP, 100x2.1mm, 2.6μm Inj vol 25 μL A) MeOH:H₂O:AcOH (20:79.9:0.1) B) MeOH:H₂O:AcOH (79:19.9:0.1) 20% B for 4 min, 20-55% B in 6 min, keep constant for 15 min, 55-100% in 5 min, keep constant for 10 min, initial conditions for 20 min 		IT GF 45 a.u.; AGF 10 a.u.; CoV 4.5 kV; CaV 40 V; ST 260 °C
3)	cts			Flow: 0.15 mL/min	Time: t_{an} =40 min, t_{Tot} =60 min T3 RP 100 x 2.1 mm, 1.7 μ m,	LOQ: 25 µg/kg all FBs
(Va clav	B_1, B_2, B_3	5	1 Add 20 mL of ACN/H ₂ O/AcOH (79.5:20:0.5) for 60 min; 2 Centrifuge to 5000 rpm, 2 min; 3 Dilute 2 mL of sample with 33 mL of PBS		40°C j vol 10 μL	QTrap ST 450°C;

ikov a et al. 201 3)	an d oth er tox ins		Clean up: 1 Load the aliquot on IAC; 2 Wash with 10 mL of ultrapure H ₂ O; 3 Elute with 3 mL of MeOH, evaporate;	A) H ₂ O, B) MeOH b 5-50% B in 1 min, 50- min, initial con	CaV 4.5kV; CUR 20 a.u.; GS1 55 a.u, GS2: 55 a.u	
	Ce rea ls, nut s		4 Reconstitute in 0.5 mL of MeOH/H ₂ O (0.5% AcOH) (1:1); 6 Filter	Flow: 0.4 mL/min	Time: t_{an} =8 min, t_{Tot} =10 min	LOD: 5 µg/kg, LOQ: 10 µg/kg all FBs
(Arr oyo - Ma nza	B ₁ , B ₂ an d oth er tox ins	2	QuEChERS 1 Add 8 mL H ₂ O into test tube; 2 Shake for 10 s; 3 Add 10 mL 5% FA in ACN; 4 Shake 2 min; 5 Add 4 g MgSO ₄ , 1 g NaCl, 1 g sodium citrate, 0.5 g Na;HCit 1.5 H ₂ O; 5 Shake for 1 min; 6 Centrifuge to 4500	1.8µn Inj va A) H ₂ O, B) MeOH bo Ar 5% B for 1 min, 5-50% l min, 72-80% B in 2 min	RHD C ₁₈ , 50 x 2.1 mm, h, 35°C h 5 μ L th with 0.3% FA, 5 mM nFo B in 1 min, 50-72% B in 2 , 80-90% for 2 min, initial in 0.2 min.	QQQ GT: 500°C; CUR: 30 psi; CaV 5 kV; GS1 and GS2 50 psi
nare s et al. 201 4)	cer eal s, spe lt, ric e		rpm, 5 min; 7 Transfer 2 mL of upper layer to a vial; 8 Evaporate; 9 Reconstitute with 1 mL of MeOH/H ₂ O 50:50; 10 Filter	Flow: 0.4 mL/min	Time: $t_{an}=8$ min, $t_{Tot}=8.2$ min	$\begin{array}{c} \text{LOD:} \\ \text{B}_1 \\ 0.20, \\ \text{B}_2 \ 0.30 \\ \mu\text{g/kg} \\ \text{LOQ:} \\ \text{B}_1 \\ 0.65, \\ \text{B}_2 \ 1.01 \\ \mu\text{g/kg} \end{array}$
	B_1, B_2		1 Add 50 mL of MeOH/H ₂ O 3:1; 2 Shake for 30 min; 3Filter			QQQ CaV
(Ce ndo ya et al. 201 9)	wh eat - bas ed pro du cts	25	Clean up: 1 Precondition with 5 mL of MeOH and 5 mL MeOH/H ₂ O 3:1; 2 Load 10 mL of filtrated; 3 Wash with 8 mL of MeOH/H ₂ O 3:1, 3 mL of MeOH; 4 Elute with 14 mL of MeOH with 0.5% AcOH ; 5 Dry (N ₂ ,40°C)	Inj vo A) H ₂ O, B) MeOl 9.5% B for 2 min, 9.5-50 in 11 min, keep for 3 m	z 2.1 mm, 3.5μm, 20°C l 45 μL H both with 1% FA % B in 1 min, 50-97.5% B in, initial condition for 5 in.	3.0 kV; ST: 150 °C; DGT 200 °C; DGF: 726 L/h;

				Flow: 0.2 mL/min	Time: $t_{an}=17$ min, $t_{Tot}=22$ min	GF 109 L/h LOD 0.01 μg/kg LOQ: 0.05 μg/kg all FBs
			Products of animal origin			
(Ga zzot ti et al. 200 9)	B ₁ Bo vin e mil	10	Clean up: 1 Centrifuge to 6000 rpm, 15 min; 2 Dilute 5 mL of sample 1:1 with H ₂ O Clean up: 1 Load the dilute sample to Vicam FumoniTestTM Immunoaffinity at 1 drop/s; 2 Wash with 20 mL of PBS buffer at 5 mL/min; 3 Elute with 1.5 mL of MeOH; 4 Pass 1.5 mL of H ₂ O, collect 3 mL; 5 Evaporate 3 mL of eluate to 1 mL (40°C, N ₂)	Inj v A) H ₂ O/ACN (90:10 (0.3 Elute isostatically with	9 x 2.15 mm, 5μm, 35°C ol 10 μL) with 0.3% FA, B) ACN 3% FA) 1 75% A-25%B for 2 min, 6 B for 3 min	QQQ CaV 3.25 kV; CoV 50 V; IST 140°C; DGT 400°C LOD:
)	k			Flow: 0.30 mL/min	Time: $t_{an}=2 \min, t_{Tot}=5 \min$	0.003 μg/kg, LOQ: 0.1 μg/kg
	$\begin{array}{c} B_1\\ B_2,\\ HF\\ B_1\\ HF\\ B_2\end{array}$		1 Homogenize in 6 mL of MeOH/H ₂ O 80:20; 2 Stir for 20 min; 3 Centrifuge to 3000 rpm, 5 min; 4 Wash twice with 6 mL of hexane; 5 Evaporate aqueous phase; 6 Reconstitute with 2 mL of aqueous buffer with 2% of AcOH, 0.1% Et ₃ N (pH 3.4)	Inj v	9 x 2.15 mm, 5μm, 35°C ol 10 μL 3) ACN both with 0.3% FA	QQQ CaV 3.25 kV; ST 140 °C; GT
(Ga zzot ti et al. 201	Dia	1	Clean up:	1	0% B in 4 min, keep for 4 min, lition for 5 min	400 °C; GF 50 L/h; DGF 890 L/h
1)	Pig liv er		 1 Condition the Oasis HLB SPE cartridges with 2 mL of MeOH and 2 mL of H₂O; 2 Load the sample; 3Wash twice: first 1 mL MeOH/H₂O (05:95), then 1 mL MeOH/H₂O/AcOH (05:94:01); 4 Elute with 2 mL of MeOH; 5 Evaporate to 200 μL; 6 Reconstitute in 1mL of mobile phase of LC 	Flow: 0.30 mL/min	Time: $t_{an} = 12 \min_{t_{Tot}} t_{Tot} = 17 \min_{t_{Tot}}$	LOD: 0.05 µg/kg, LOQ: 10 µg/kg all FBs and

						analog ues
(Sør	B ₁ , B ₂		1 Add 140 μL of IS (¹³ C-FB ₂ 0.5 μg/mL), 4.5 mL of H ₂ O, 2.5 mL of ACN, 6 mL of pentane; 2 Shake for 1 h; 3 Centrifuge to 8000 g, 10 min; 4- Discard upper phase; 5 Transfer 3.5 mL of lower phase; 6 Add 9 mL of acetone; 7 Shake; 8 Centrifuge to 8000 g, 10 min; 10 Collect 100 mL upper phase; 11 Evaporate to 1.5 mL (45°C), reconstitute in 0.25 mL of MeOH	Gemini C6-phenyl 5(Inj vo A) H ₂ O, B) ACN b	11μL	QQQ IST 120°C; DGF
ense n, Mo		0		20-55% B in 6 min, then 2.5 m		700L/h ; DGT 350°C
gen sen, and Niel sen	Me at pro du	0. 7	Clean up: 1 Load sample in Oasis (MAX) SPE cartridges; 2condition with 1 mL of MeOH followed by 1 mL of H ₂ O, wash with 1 mL of 1% aqueous ammonia; 1 mL of MeOH/H ₂ O/HCl 37% (40:59:1); 3 elute with 2 mL of 2% AcOH in MeOH; 4 evaporate (N ₂ , 45°C), re-dissolve in 200 µL ACN/H ₂ O (1:2).			LOD: B ₁ 64 µg/kg, B ₂ 6
201 0)	cts			Flow: 0.30 mL/min	Time: <i>t</i> _{Tot} =9 min	$\mu g/kg,$ LOQ: B ₁ and B ₂ 150 $\mu g/kg$
	B_1, B_2		1 Filter, 2 Dilute 1:1 with 10 mL of PBS, 3 Mix for 3 min			QQQ CaV
(Lili ana J.G. Silv a et al.	Uri ne	10 m L	Clean up: 1 Load the sample in FumoniTestTM immunoaffinity column; 2 Wash with 10 mL PBS; 3 Elute with 5 mL of MeOH; 4 Dry (N ₂ , 60°C); 5 Redissolve in 1 mL of MeOH/H ₂ O (1:1)	Inj vol A) H ₂ O, B) MeOH 65% B for 3 min, 65-759	Luna C ₁₈ , 150 x 4.6mm, 5μm, 30°C Inj vol 20 μL A) H ₂ O, B) MeOH both with 0.5% FA 65% B for 3 min, 65-75% B in 4 min, keep for 8 min, initial condition for 10 min	
201 0)				Flow: 0.50 mL/min	Time: t_{an} =15 min, t_{Tot} =25 min	LOD: 5 µg/L LOQ: 10 µg/L all FBs
(Šar kanj et	B ₁ an d oth er tox ins	50 0	1 Centrifuge to 5600 g, 3 min; 2 Incubate with 500 μL PBS (200 mM, pH 7.4) containing 3000 U of β- glucuronidase, 16 h, 37 °C	Acquity HSS T3, 100 : Inj vol A) H ₂ O, B) ACN, bc 10% B for 2 min, 10-50%	10 μL oth with 0.1% AcOH	Qtrap ISV 4.50 kV; ST 550°C; CUR 30 psi;
al. 201 8)	Uri ne	μ L	Clean up: 1 Precondition with 1mL MeOH, 1mL H ₂ O; 2 Add sample to Oasis PRiME HLB; 3 Wash twice with 500 μL H ₂ O; 4 Eluate with 200 μL ACN x 3; 5 Evaporated (N ₂); 6 Reconstitute with 470 μL of 10% ACN,	in 5 min, hold 4 min, ini	itial condition for 3 min.	SG 80 psi; DG 80 psi
			0.1% AcOH, add 30 μL IS (0.38 ng/mL ¹³ C-FB ₁)	Flow: 0.1 mL/min	Time: $t_{an} = 24 \text{ min},$ $t_{Tot} = 27 \text{ min}$	LOD: 0.001

						/1
				66		μg/L, LOQ: 0.01 μg/L
			Bevearages			
(Na kag awa et al. 202	B ₁ , B ₂ , B ₃	5 m L	1 Add 0.1 mL of IS (¹³ C ₃₄ -FB1 0.2 mg/L in acetonitrile: water (1:1); 2 adjusted volume at 10 mL with wine; 3 mix; 4 add 8mL of PBS (1% PEG, 5% NaHCO ₃ ; 5 mix;	ZORBAX Eclipse XDB-C ₁ Inj vol 3- A) H ₂ O, B) ACN, b 10% B for 3 min, 10-90% min, initial condit	-20 μL oth with 0.1% FA B in 15 min, hold for 5	Qtrap CaV 5 kV; CUR 10 psi; GS1 70 psi; SG 60 psi; ST 500°C
0)	Do me stic wi ne		Clean up: 1 Equilibrate with 3 mL of PBS; 2 Load sample in cartridge; 3 Wash 6 mL (3 mL x 2 times) of H ₂ O (0.5% NaHCO ₃) and 6 mL (3 mL x 2 times) of 10 mM AmAc; 4 Elute with 3 mL of MeOH (2% AcOH); 5 evaporate to dryness; 6 reconstitute in 0.2 mL ACN:H ₂ O 1:1.	Flow: 0.3 mL/min	Time: $t_{an} = 20$ min, $t_{Tot} = 30$ min	LOD: 1 µg/Kg all Fbs LOQ 2
(Ro mer o- Gon zále	B ₁ , B ₂	10 m L	1 Sonicate for 20 min	Acquity UPLC BEH C ₁₈ , 30° A) H2O, B) MeOH bo 25 to 100% B in 3.75 min 25% B in 0.5 min, initia	C th with 5 mM AmFo , keep 1.25 min, 100 to	QQQ CaV 3.5 kV; ST 120; DGT 350°C; CGF 80 L/h; DGF 600 L/min
z et al. 200 9)	Be er		1 Precondition with 5 mL ACN/H ₂ O (60:40) and 5 mL of H ₂ O; 2 Load sample in C ₁₈ cartridge; 3 Wash with 5 mL of H ₂ O; 4 Elute 2 mL ACN/MeOH 60:40; 5Filter	Flow: 0.35 mL/min	Time: t_{an} = 5.5 min, t_{Tot} = 6.5 min	LOD: B_1 0.07, B_2 0.09 $\mu g/kg;$ LOQ: B_1 0.23, B_2 0.30 $\mu g/kg$
(Ta mzu ra, Uya ma,	$\begin{array}{c} B_1,\\ B_2,\\ B_3,\\ an\\ d\end{array}$	10 m L	1 Sonicate for 15 min, 2 Add 10 mL ACN, mix, 3 Add the content of dSPE citrate extraction tube; 4 Vortex for 20 s, 5 Centrifuge to 2380 g, 5 min	Acquity UPLC BEH C ₁₈ 40° Inj vol A) H ₂ O, B) MeOH (2% A 55-80% B	C 5 μL AcOH, 0.1 mM AmAc)	QQQ CaV 3 kV; IST 120°C;

and Mo chiz uki 201 1)	oth er tox ins Be er- bas ed dri nks	-	Clean up: 1 Precondition with 5 mL ACN; 2 Load sample in InertSep C ₁₈ , SPE; 3 Elute with 5 mL ACN; 4 Evaporate to dryness; 4 Dissolved with 500 μL of 10 mM AmAc aqueous/ACN (85:15); 6 Filter	Flow: 0.50 mL/min	Time: $t_{an} = t_{Tot} = 2 \min$	DGT 450°C; CGF 50 L/h; DGF 800 L/h LOQ: 5 μg/L all FBs
			Other samples			
	$\begin{array}{c} B_1,\\ B_2,\\ B_3\\ an\\ d\\ oth\\ er\\ tox\\ ins \end{array}$		 1 Add 25 mL of AcOEt/FA 95:5 for 30 min ; 2 Centrifuge; 3 Evaporate 20 mL to dryness; 4 Reconstitute in 5 mL of H₂O/MeOH 1:1 and 10 mL Hex; 5 Shake, 6 Transfer aqueous fraction into a tube; 7 Add H₂O/MeOH 1:1 (2 x 5mL); 8 Evaporate; 9 Reconstitute in 400 μL H₂O/MeOH 1:1; 10 Centrifuge to 14000 g, 10 min; 12 Take 250 μL, 13 Filter; 14 Dilute in 25 mL H₂O 	Symmetry C ₁₈ , 150 x Inj vol 2 A) H ₂ O/MeOH/A MeOH/H ₂ O/AcOH 97:2:1 5-65 % B in 7 min, 65-75% in 2 min, keep for 2 min, 1	20 μL cOH 94:5:1, B) both with 5mM AmAc 6 B in 4 min, 75-100% B	QQQ CaV 3.2 kV; ST 150°C; DGT 350°C; CGF
(di Ma vun gu et al. 200 9)	Fo od	1	Clean up: SPE 1 Condition with 10 mL CH ₂ Cl ₂ /MeOH 8:2 with 50 mM FA, then 5 mL MeOH, 20 mL acidified H ₂ O (10	40% B in 6 min, 40-5% I	B in 1 min, hold 2 min.	20 L/h; DGF50 0 L/h LOD: B ₁ 1,
9)	sup ple me nts		 mM HCl), finally 10 mL H₂O; 2 Add obtained solution to Oasis HLB SPE cartridge; 3 Wash 10 mL H₂O; 4 Elute with 1 mL MeOH and 4 mL CH₂Cl₂/MeOH 8:2; 5 Evaporate; 6. Reconstitute in 100 μL injection solvent; 7 centrifuge 14000g for 10 min. 	Flow: 0.3 mL/min	Time: t_{an} = 16 min, t_{Tot} = 25 min	FB ₂ 0.3, FB ₃ 1 μg/kg LOQ: B ₁ 3, FB ₂ 1, FB ₃ 3 μg/kg
	$\begin{array}{c} B_1,\\ B_2\end{array}$		Add 40 mL ACN/H ₂ O (1:1); 2 Shake 5 min; 3Filter	Inertsil ODS, 350 x 2	· · ·	QQQ CaV 4 kV;
(Kh ayo on et al. 201	Fo od, fee	10	Clean up: 1 Take 1 mL of filtrate; 2 Add 2.5 mL of 1% KCl; 2 Precondition with 5 mL of MeOH, follow of 5 mL 1% KCl solution; 3 Load in C ₁₈ , SPE; 4 wash with 3 ml 1% KCl, followed by 2 mL of ACN/1% KCl 1:9;	Inj vol 2 A) H ₂ O, B) MeOH t 50-75% B in 4.0 min, 75-1 min, B 100-5	both with 0.2% FA 00% in 2.0 min, keep 6.5	DGF 600 L/h; DGT 350°C
0)	d		5 Elute with 2 mL of MeOH/H ₂ O (1:1)	Flow: 12.5 min 0.20 mL/min, 3 min 0.3 mL/min	Time: t_{an} =12.5, t_{Tot} =15.5 min	LOD: B ₁ 10,

				<u>k</u> C		$\begin{array}{c} B_2 40 \\ \mu g/kg \\ LOQ: \\ B_1 40, \\ B_2 130 \\ \mu g/kg \end{array}$
(Jer ome Jeya kum	$\begin{array}{c} B_1, \\ B_2 \\ an \\ d \\ oth \\ er \\ tox \\ ins \\ Fu \end{array}$	-	1 Add 25 mL AcOEt to cultures, shake to 8000 rpm; 2 After 2 h, mix with 5% acetone, isopropanol; 3 Extracted with AcOEt 1:1; 4 Collect upper layer, 6 Evaporate; 7 Reconstitute 10 mL isopropanol	Supelco C₁₈, 250 Inj vol A) H ₂ O, 0.1% 15% B, 5 min, 15-100% E 100-15 % B in 1 m	10 μL FA, B) ACN in 35 min, keep 10 min;	Qtrap CaV 5 kV; ST 200°C; DGT 300°C, NGF
ar, Zha ng, and Thir uve nga dam 201 8)	ru ng al cul tur es: Ma ize , As par ag us	N R	Clean up: SAX 1 Add 10 mL sample into cartridge; 2 Eluate 3 mL MeOH followed by 5 mL of 1% KCl; 3 Collect into a 5-mL tube; 4 Dry	Flow: 0.2 mL/min	Time: t _{an} =50 min, t _{Tot} =60 min	2μmL/ min LOD/L OQ: NR
(Fac orro	B_1, B_2		QuEChERS. 1 Add 10 mL of ACN/FA 90:10; 2 Shake for 1 h, 25°C; 3 Add 0.5 Na citrate sesquihydrate+1g NaCitrate+1g NaCl+4g MgSO4; 4 Shake for 1 min; 5 Centrifuge to 3398 g, 5 min	Kinetex C ₁₈ , 50 x 2.1 Inj vol		QTOF CaV 5.5 kV;
Llo mpa rt, and	Mi xe d Fe	2	Clean up: 1 Discard of supernatant; 2 Load 1 mL in SPE Oasis PRiME HLB cartridge (3cc, 150 mg), collect; 3 Transfer to a 2 mL dSPE tube; 4 Add 150 mg MgSO ₄ +50 mg PSA+30 mg C ₁₈ silica+30 mg Al-N; 4	a) H ₂ O, B) MeOH, both bu or An 10%-100% B in 8	iffered with 3 mM AmFo nAc.	ST 550 °C, CUR 50 a.u.
Dag nac 202 0)	ed Rat ion s		Centrifuge to 2360 g, 2 min; 5 Take 500 µL, evaporate; 6 Reconstitute with 350 µL of MeOH	Flow: 0.25mL/min	Time: $t_{an} = t_{Tot} = 15 \text{ min}$	LOQ: B ₁ 2.9, B ₂ 2.4 μg/L
(Jia et al. 201 4)	$\begin{array}{c} B_1,\\ B_2,\\ B_3\\ an\\ d\\ oth\\ er\\ tox\\ ins \end{array}$	15	1 Add 10 mL MeOH/H ₂ O (84:16) with 1% AcOH; 2 Vortex 1 min, add 6 g MgSO ₄ +1.45 g sodium acetate anhydrous; 3 Shake for 1 min; 4 Centrifuge to 4000 rpm, 5 min	Thermo Accucore C ₁₈ , Inj vol A) H ₂ O, B) MeOH, both 0% B for 1 min, 0-100% 100-0% B in 1 min, init	5 μL 0.1% FA, 4 mM AmFo B in 6 min, keep 5 min,	Q- Orbitr ap CaV 3kV; ST 320 °C; GT 350 °C; SG

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	Da iry pro du		Clean up: 1 Add 8 mL of upper phase+1.2 g MgSO ₄ +108 mg PSA+405 mg C ₁₈ silica to dSPE tube; 2 Shake for 1 min; 3 Centrifuge to 4000 rpm, 5 min; 4 Transfer 200 μL; 5 Add 300 μL of MeOH+500 μL 8 mM AmFo; 6	ξÇ)	18 L/min, Aux 3 L/min
	cts		Vortex 30 s; 7 Filter 1 mL	Flow: 0.30 mL/min	Time: $t_{an}=12 \text{ min},$ $t_{Tot}=15 \text{ min}$	LOD/ LOQ: NR
	B ₁ , B ₂ , B ₃ an d oth er tox ins	-	1 Add 15 mL AcOEt, FA 1%); 2 Shake 15 min; 3 Centrifuge to 3300 g, 5 min; 4 Filtrate; 5 Repeat this process with 10 mL of the same mix solvent; 6 Keep an aliquot (10 mL) for the SAX; 7 Evaporate remaining part to 5 mL;	Inj v A) H ₂ O/MeOH MeOH/H ₂ O/AcOH (97:	x 2.1 mm, 5 μm, 25 °C ol 20 μL /AcOH (94:5:1). B) 2:1) both with 5 mM AmAc 25% B in 4 min, 100% B for	QQQ CaV: 3.2 kV, ST: 150
(Mo nbal iu et al. 200	Sw	3	Clean up:	2 min, initial co	onditions for 12 min	°C, DGT: 350 °C
9)	eet pe pp er		 SPE: pass remaining through the NH₂-SPE column; 2 evaporate; 3 Redissolve the evaporate in 3 mL of ACN/H₂O (84:16); 4 Pass through the SPE; 3SAX, evaporate aliquot to dry; 5 Redissolve in 5 mL MeOH/H₂O (75:25); 5 Adjust pH at 5.8-6 with NaOH 0.25 M; 6 Wash with 4 mL MeOH/H₂O (75:25) and then 4 mL of MeOH; 7 Elute with 4 mL MeOH, AcOH 1%; 8 Evaporate; 9 Redissolve in 100 µL H₂O:MeOH:AcOH (57.2:41.8:1) and 5 mM of AmAc; 10 Centrifuge to 14000 g, 15 min 	Flow: 0.30 mL/min	Time: t_{an} =13 min, t_{Tot} =25 min	$\begin{array}{c} B_{1} \ 13, \\ FB_{2} \\ 6.5, \ B_{3} \\ 8.4 \\ \mu g/kg \\ LOQ: \\ B_{1} \ 27, \\ FB_{2} \\ 13, \ B_{3} \\ 17 \end{array}$
(de Sme t et	$\begin{array}{c} B_1, \\ B_2, \\ B_3 \\ \hline Bel \\ l \\ pe \\ pp \end{array}$	1	1 Add 8 mL of ACN/H ₂ O 84:16; 2 Shaker 30 min; 3Centrifuged to 2670 g, 20 min; 4 Evaporate Clean up:	Inj v H ₂ O/ACN (60	50 x 3.2 mm, 5μm ol 20 μL :40) with 0.3% FA ic condition	μg/kg QQQ CaV 3.6 eV; ST 140 °C; DGT 350 °C
al. 200 9)	er, ric e, cor n fla		1 Condition with 2 mL MeOH; 2 Wash with MeOH/H ₂ O 75:25; 3 Redissolve sample in 2 mL of MeOH/H ₂ O (75:25); 4 Adjust pH 5.8-6.5 with 0.1M NaOH; 5 Eluate with 2 mL MeOH/FA 95:5; 6 Evaporate; 7 Redissolve in 100 μL of H ₂ OACN 60:40 with 0.3% FA	Flow: 0.3 mL/min	Time: 12 min	LOD: B ₁ 20, B ₂ 7.5, B ₃ 12.5 µg/kg

kes C	LOQ: B ₁ 40, B ₂ 15, B ₃ 25 µg/kg
(ACN) Acetonitrile, (AcOH) Acetic acid, (AE) Appearance energy, (AmAc): ammonium acetate, (AmFo) Ammonium formate, (CaV) Capillary voltage, (CaT) Capillary tem	
gas flow, (CoG) Collision gas, (CUR) Curtain gas, (DG) Drying gas, (DGF) Desolvation gas flow, (DGT) Desolvation gas temperature, (EV) Extractor voltage, (FA) Form	ic acid, (Frag Vol)
Fragmentor Voltage, (GF) Gas flow, (GT) Gas Temp, (LIT) linear ion tramp, (MeOH) Methanol, (MSPD) Matrix Solid Phase Dispertion, (NG) Nebulizer gas, (NR) Not repo	
Liquid Extraction, (RT) Room temperature, (ST) Source temperature, (SV) Source voltage, (tan) analysis time, (trot) total time including column conditioning	g.

1 460 00 01 //

	Table 3. LC methods with different detectors								
[R ef]	F B S M at ri x	S a pl e (g)	Sample treatment	LC c	Detecto r conditi ons, Limits				
	·		Maize and corn-based products	U					
(Wa ll- Mar tíne z et al. 201 9)	B ₁ , B ₂ an d ot he r to xi ns <i>To</i> <i>rti</i> <i>lla</i>	25	 1 Dry the <i>tortilla</i> at 60°C for 2.5 h; 2 Milled and homogenize for 15 min at 30 rpm; 3 Add 50 mL MeOH/H₂O 80:20; 4 Shake for 2 min; 5 Centrifuge to 4000 rpm, 10 min; 6 Take 10 mL of supernatant; 7 Dilute adding 40 mL of PBS Clean up: IAC R-Biopharm 1 Precondition with 20 mL PBS (5.0 mL/min); 2 Load 10 mL of sample diluted on the cartridge; 3 Wash with 1.5 mL MeOH (0.5-1.0 mL/min) and 1.5 mL of H₂O Mix 100 µL of diluted extract with 100 µL OPA reagent (120 mg OPA, 3mL MeOH, 12 mL Na₂B₄O₇· H₂O 0.1 M, 179 µL 2-mercaptoehtanol) prior to injection. 	5µ1 Iny v A) 99% H ₂ O, B) A(41 % B 9 min, 61 %	DB, ODS, 250 x 4.6 mm, m, 40°C vol 10 μ L CN both with 1 % AcOH B for 7 min, keep 4 min, litions for 5 min Time _a t _{an} = 20 min, t _{Tot} =25 min	FDA λex : 360 nm λem: 450 nm LOD: B ₁ 0.13, B ₂ 0.04 μg/kg LOQ: B ₁ 3.0, B ₂			
(Cal das and Silv a 200 7)	B ₁ , B ₂ an d ot he r to xi ns Co	25 co rn m eal , pr ec oo ke d co rn	1 Add 100 mL MeOH/H ₂ O (3:1) (cornmeal, PCF, popcorn, corn snacks), 50 mLMeOH/H ₂ O (4:1) + 2.5 g NCl (sweet corn) 100 mL MeOH:0.4 M sodium tetraborate (3:1) (corn flakes), 2 filter Clean up: SAX 1 Precondition with 5mL MeOH:H ₂ O 1:1; 2 Add 10 mL of filtrate on SAX column; 3 Wash with 5mL	Iny A) H ₂ O, B) AC 55-80% B in 5 min, ke	n x 4.6 mm, NR vol 10μL N both 2.5% AcOH æp 8 min, initial conditions r 1 min	FDA λex : 420 nm λem: 500 nm			

	se fo	ur, po	popcorn, corn snacks), 12 mL MeOH/AcOH (99:1) + 8 mL MeOH:AcOH (95:5) (sweet corn, corn flakes); 4 Dry at 40°C; 5 Reconstitute in 500 μL			
	od pr od uc ts	pc or n, sw eet co rn, co rn fla ke s 12 .5 co rn sn ac ks	Derivatization 1 Add 480 μ L of 0.05 M sodium borate buffer (pH 9.5), 170 μ L sodium cyanide solution (0.013%) and 50 μ L of 0.5 mg/mL naphthalene-2,3 dicarboxaldehyde (NDA) in MeOH; 2 Vortex, 3 Heat at 60 °C for 15 min, and cooled, 4 Add 2.8 mL of 0.05 M phosphate buffer	Flow: 1 mL/min	Time: $t_{an} = 13 \min t_{Tot} = 14 \min$	LOQ: 127-2040 µg/kg depending on the matrix
(Chi ara Dall 'Ast	$\begin{array}{c} B_1\\\B_2\\\B_3\end{array}$		1 Add 100 mL H ₂ O/ACN 1:1; 2 Shake for 1h; 3 Filter; 4 Adjust to pH 6-9 with 0.5 N NaOH; 4Take 3 mL, place into test tube; 5 Add 8 mL MeOH/H ₂ O 3:1 Clean up: SPE MultiSep 211 Fum	Inj v H ₂ O/ACN	100 x 4.6, 5μm, NR ol 80 μL /AcOH 52:47:1	FDA λex: 420 nm λem: 500
a, Ma ngia , et	Gr ou nd		1 Precondition with 5 mL MeOH, then 5 mL MeOH/H ₂ O 3:1; 2 Load sample 3Wash with 8 mL MeOH/H ₂ O, then 3 mL MeOH; 4Elute 3:1 MeOH/AcOH 99:1; 5 Dry at the eluate to 60°C; 6 Reconstitute with 1 mL MeOH	is	ocratic	nm
, et al. 200 9)	co rn		Derivatization 1 Add 1 mL sodium borate buffer (0.05 M, pH 9.5), 0.5 mL NaCN reagent (13 mg NaCN in 100 mL water) and 0.5 mL NDA reagent (25 mg NDA in 100 mL MeOH). 2 Heat for 20 min in a 60°C water bath and cooling for 4 min at 8°C, 3 Dilute with 7 mL of phosphate buffer (0.05 M, pH 7.4)/ACN (2:3)	Flow: 0.2 mL/min	Time _{tot} : NR	LOD/LO Q: < 100 µg/kg
	B_1 , B_2		1 Add 40 mL MeOH/H ₂ O 4:1; 2 Centrifuge 2500 g, 15 min; 3 Extract remaining solid twice with 30 mL MeOH/H ₂ O 4:1; 4 Filter		250 x 4.6mm, 5μm, NR vol NR	FDA λex: 420
(L. J.G. Silv a et	M aiz e	25	Clean up: FumoniTest TM IAC 1 Dilute 10 mL sample with 40 mL PBS; 2 Filter; 3 Load 20 mL; 3 Wash 10 mL PBS; 5 Elute 2 x 1.5 mL MeOH; 6 Evaporated at 60°C	ACN/H ₂ O	AcOH 61:38:1 ocratic	nm λem: 500 nm
a et al. 200 7)	ba se pr od uc ts		Derivatization 1 Reconstitute in 50 µL MeOH/H ₂ O 1:1; 2 Add 500 µL 0.05M sodium borate buffer (pH 9.5 adjusted with 1N NaOH), 500 µL NaCN reagent, and 150 µL NDA reagent (0.5 mg/mL in ACN); 2Heat 15 min at 60°C, cold to room temp	Flow: 1 mL/min	Time _{tot} : 14 min	LOD/LO Q: NR

B_1	5	4 Evaporate; 5 Dry, 40°C; 6 Reconstitute in 0.5mL of MeOH/0.1M phosphate buffer at pH 3.15 3:2 Derivatization: OPA NR	Flow: 0.8 mL/min	Time: $t_{an} = 10 \min t_{Tot} = 15 \min$	LOD B ₁ 4, B ₂ 5 µg/L LOQ: B ₁
,		1. Hedrete for 12 havids 10 mJ advances H.O. 2. Add 20 mJ ACN 2. Challs 120 11. 4. Fill			13, B ₂ 16 μg/L **
		1 Hydrate for 12 h with 10 mL ultrapure H ₂ O; 2 Add 30 mL ACN, 3 Shake 120 rpm, 1 h; 4Filter	Agilent C ₁₈ , 250 x 4.6 mm, 5μm, 40°C Inj vol 50 μL		FDA λex: 335
M	10	Clean up: SAX 1 Precondition with 5 mL MeOH followed by 5 mL ACN /H ₂ O 3:1 at a flow rate 1 mL/min; 2 Load 8 mL of sample; 3 Wash with 5 mL MeOH; 4 Elute with 10 mL AcOH/MeOH 1:99, 5Dry, 6 Reconstitute in 2 mL ACN/H ₂ O 1:1	55-65% B in 10 min, 6	buffer (pH = 4), B) MeOH 5-70% B in 12 min, keep 3 5% B in 3 min	nm λem: 440 nm
iz		Derivatization: OPA post column in HPLC pump 1 (1% NaOH) 0.8 mL/min, pump 2 (3g OPA, 9 mL MeOH, potassium borate, 9 mL 2-mercaptoethanol) 0.6 mL/min	Flow: 1 mL/min	Time: $t_{an} = 25 \min t_{Tot} = 28 \min t_{Tot}$	LOD: B ₁ 6, B ₂ 7 µg/kg LOQ: B ₁ 20, B ₂ 23 µg/kg
	20	1 Add 100 mL of MeOH:H ₂ O 7:3; 2 shake for 3 min; 3 filter; 4 dilute 1:20 with deionized H ₂ O; 5 derivatized with OPA	Vol. A) 20 % H ₂ O (1% N	. Inj: NR Na ₃ PO ₄) B) 80% MeOH	FDA λex: 335 nm λem: 440 nm
M liz e			Flow: 1 mL/min	t _{Tot} = 30 min	LOD: 223 µg/kg for all FBs
B ₁ an d ot ne r to xi ns	10	1 Add 50 mL of MeOH/H ₂ O 75:25; 2 Mix for 1 min; 3 Filter	Inj v MeOH/0.1 M sodium d adjust pH 3.35 v	rol 10 μL lihydrogen phosphate 80:20 with phosphoric acid	
B_1 B_2 B_1 B_2 B_1 B_1 B_1 B_1 B_1 B_1 B_1 B_1 B_1 B_1 B_2 B_1 B_1 B_2 B_1 B_2 B_1 B_1 B_2 B_1 B_1 B_2 B_1 B_1 B_1 B_2 B_1 B_1 B_1 B_2 B_1		20	10 mL ACN/H2O 1:1 Derivatization: OPA post column in HPLC pump 1 (1% NaOH) 0.8 mL/min, pump 2 (3g OPA, 9 mL MeOH, potassium borate, 9 mL 2-mercaptoethanol) 0.6 mL/min 20 1 Add 100 mL of MeOH:H2O 7:3; 2 shake for 3 min; 3 filter; 4 dilute 1:20 with deionized H2O; 5 derivatized with OPA 10 1 Add 50 mL of MeOH/H2O 75:25; 2 Mix for 1 min; 3 Filter	10 mL ACN/H ₂ O 1:1 Derivatization: OPA post column in HPLC pump 1 (1% NaOH) 0.8 mL/min, pump 2 (3g OPA, 9 mL MeOH, potassium borate, 9 mL 2-mercaptoethanol) 0.6 mL/min Flow: 1 mL/min 20 1 Add 100 mL of MeOH:H ₂ O 7:3; 2 shake for 3 min; 3 filter; 4 dilute 1:20 with deionized H ₂ O; 5 derivatized with OPA Zorbax Eclipse 0 Vol A) 20 % H ₂ O (1% 1) isocra 10 1 Add 100 mL of MeOH:H ₂ O 7:3; 2 shake for 3 min; 3 filter; 4 dilute 1:20 with deionized H ₂ O; 5 derivatized with OPA Flow: 1 mL/min 10 1 Add 50 mL of MeOH/H ₂ O 75:25; 2 Mix for 1 min; 3 Filter Supercosil C ₁₈ , 15 10 1 Add 50 mL of MeOH/H ₂ O 75:25; 2 Mix for 1 min; 3 Filter Supercosil C ₁₈ , 15 10 1 Add 50 mL of MeOH/H ₂ O 75:25; 2 Mix for 1 min; 3 Filter Supercosil C ₁₈ , 15 10 1 Add 50 mL of MeOH/H ₂ O 75:25; 2 Mix for 1 min; 3 Filter Supercosil C ₁₈ , 15	$ \begin{array}{c c c c c c c c } \hline 10 & \underline{\text{mL ACN/H}_{2}O 1:1} \\ \hline \hline \text{Derivatization: OPA post column in HPLC pump 1 (1% NaOH) 0.8 mL/min, pump 2 (3g OPA, 9 mL MeOH, potassium borate, 9 mL 2-mercaptoethanol) 0.6 mL/min & Flow: 1 mL/min & Time: t_{ws}= 25 min t_{Tot}= 28 min \\ \hline 28 min & \underline{28 min} & $

			Clean up: SAX 1 Precondition with 5 mL H ₂ O, 5 mL			
			MeOH, and 5 mL MeOH/H ₂ O 75:25 at a flow rate of 1 mL/min; 2 Apply 10 mL of sample; 3 Wash 8 mL			
	Ch ips		with 5 mL MeOH/H ₂ O 75:25 and 8 mL MeOH; 4 Elute 14 mL MeOH/AcOH 99:1 at a flow 1 mL/min; 5			
			Evaporate; 6 Reconstitute in 1 mL of MeOH, 7Evaporate, 8Reconstitute in 200 µL MeOH			LOD
			Derivatization: OPA 1 Mix 50 µL sample with 200 µL OPA (40 mg OPA in 1 ml of methanol	Flow: 1 mL/min	Time: NR	LOD: 0.025
			followed by addition of 5 ml of 0.L M sodium borate solution and 50 μ L of 2-mercaptoethanol.)	TIOW. TIME/IIII	Time. Nix	μg/kg
		1 Homogenize in 2 mL of distilled H ₂ O (Liver, 500 rpm; breast muscle 3000rpm, 20 s) with a teflon Potter,				
	B ₁		2 Precipitate proteins with 2 mL of MeCN/MeOH 1:1 and 25mg of NaCl; 3Stir to 300 rpm, 120 min; 4			
1			Centrifuge 3000 g, 15 min; 5Take 3 mL of supernatant; 6Add 4 mL Hex; 7 Centrifuge 3000 g, 15 min; Prontosil C ₁₈ , 250 x 4.6mm, 5µm		· ·	FDA
			8 Take 2 mL aqueous phase; 9 Dilute with 8 mL PBS Clean up: IAC FUMONIPREPC	Inj v	1 225	
	A ni		1 Pass the sample through cartridge; 2 Wash 10 mL of PBS (pH 7.4); 3Elute 1.5 mL of MeOH, 1.5mL of	MeOH/NaH ₂ PO ₄ 0.1M pH 3.35, 75:25 Isocratic		λex : 335 λem: 440
(Tar	m		H ₂ O; 4			
dieu	al		Evaporate, 40°C; 5Reconstitute with 200 µL ACN/H ₂ O 1:1			
et al.	tis					
200	su es, liv		Derivatization: OPA			
8)						LOD: 10
	er,		1 Add to 50 μ L of sample: 50 μ L of OPA, 50 μ L of 0.1M borate buffer at pH 8.3, and 50 μ L of H ₂ O	Flow: 1 mL/min	Time _{tot} : 15 min	μg/kg LOQ: 13
	ki					μg/kg
	dn					100
	ey					
(Ka	B ₁ an					
was	d			Spherisorb ODS-2, 250 x 4.6 mm 2, 5μm Inj vol 20 μL ACN/H ₂ O/AcOH 54:46:1 Isocratic		
him	ot		Clean up: SAX			
a,	he	>5	1 Adjust pH 5.8-6.5 with 1N NaOH; 2 Filter; 3 Precondition with 10 mL of MeOH, 10 mL MeOH/H ₂ O 3:1; 4 Load 50 mL; 5 Apply into SAX; 6 Wash with 10 mL MeOH/H ₂ O 3:1 and 6 mL MeOH; 7 Elute			FDA λex : 335
Viei	r	0	with 20 mL MeOH/AcOH 95:5; 8Dry with N_2 at 60°C ACN/H ₂ O/AcOH 54:46:1 Isocratic			$\lambda em: 440$
ra,	to	m L				
and Val	xi ns	L N				
ente	115	R		-		
Soa	Be					
res	er		ftaldialdehyde in 1 mL ethanol diluted with 0.1 M borate buffer and 50 µL 2-mercaptoethanol; 4. Ultrasonic	Flow: 1 mL/min	Time _{tot} : 19 min	LOD/LO
200			bath at 5-15 °C, 30 sec	110W. 1 IIIL/IIIII	Timetot. Ty imm	Q: NR
7)	B ₁	N				
(Jer		R				
ome	B ₂	>5				
Jeya	an	0		C ₁₈ , 250 x 4.6 mm, 5μm		FDA
kum	d	m	1 Add 25 mL AcOEt to the culture; 2 Shake to 8000 rpm; 3 Filtrate after 2h; 4 Mix with 5% acetone,	Inj vol 20 μL		λex: 335
ar, Zha	ot he	L	L isopropanol; 5Extract liquid phase with AcOEt in a 1:1 ratio; 6 Collect upper phase, 7- Evaporate; 8 Reconstitute in isopropanol. All x 2		1 M sodium dihydrogen	nm
ng,	r		Reconstitute in isopropanol. All x 3 phosphate buffer pH 3.3; 75:25			λem: 440 nm
and	to					11111
Thir	xi					
uve	ns					

nga dam 201	Su ga		Clean up: SAX 1 Load 10 mL; 2 Wash with 3 mL MeOH followed by 5 mL of 1% KCl; 3 Collect into 5 mL tube; 4 Evaporate	6.0		
8)	rc an e		Derivatization: OPA 1. Pre-column derivatization [50 mg of OPA in 1.25 mL of methanol + 50 µL of 2-mercaptoethanol + 11.2 mL of 0.1 M sodium borate buffer (pH 9.5)]; 2 Mix 100 µL of sample with 25 µL of OPA; 3incubate 2 min to rt	Flow: 0.3 mL/min	Time: 16 min	LOD/LO Q:NR
(Pia cent ini et al. 201 7)	B ₁ an d ot he r to xi ns	>5 0 m	Clean up 1 Adjust pH 5.8-6.5 with 1N NaOH; 2 Filter; 3 Precondition with 10 mL of MeOH, 10 mL MeOH/H ₂ O 3:1; 4 Load 50 mL; 5 Apply into SAX; 6 Wash with 10 mL MeOH/H ₂ O 3:1 and 6 mL MeOH; 7 Elute with 5 mL MeOH/AcOH 95:5; 8Dry with N ₂ at 60°C, 7 Reconstitute in 300 µL ACN/H ₂ O 1:1, 8Filter	Luna C ₁₈ , 150 x 4.60 mm, 5 μm, Temp NR Inj vol 20 μL ACN/H ₂ O/AcOH 520:480:5 Isocratic		FDA λex: 335 nm λem:440 nm
	Be er	L 25	Derivatization: OPA 1Take 500 µL; 2Add 200 µL OPA reagent (40 mg <i>O</i> -ftaldialdehyde in 1 mL ethanol diluted with 5 mL 0.1 M borate buffer and 50 µL 2-mercaptoethanol			LOD: 2 µg/L, LOQ: 6.3
	$\begin{array}{c} B_1\\ ,\\ B_2\end{array}$		1 Add 50 mL MeOH/ACN/H ₂ O 1:1:2; 2Vortex for 30 s; 3Shake 20 min, 3Filter, 4Dilute 1:5 with 0.01 M PBS	Acclaim 120 C ₁₈ , 4.6 x 150 mm; 3 μ, 35°C 10 μL of sample A) citrate buffer (pH 4.7): ACN (70:30)], 20% B		μg/L FDA λex: 263
(Sm ith	Fe	25 0.0 10 1 Tak (125 µL and d	Clean up: SPE 1 Take 5 mL aliquot; 2apply into SPE column; elute rate approximately 1-2 drops/s, 3Wash with 10 mL 0.01 M PBS, 4Removed solvent (vacuum, 5 min), 5Elute with 1.5 mL MeOH then 1.5 mL H ₂ O	[citrate buffer (pH 4.7): ACN (30:70) 20-95% B in 20 min post-injection, keep 5 min, 95-20 % B in 1 min, initial conditions for 4 min		nm λem: 313 nm
et al. 201 7)			Derivatization: Fmoc 1 Take 500 μL, 2- Add boric acid (1 M, pH 7.5, 125 μL), control pH during derivatization; 3 Add Fmoc (125 μL, 0.12 g Fmoc, 40 mL ACN, 2.88 g citric acid, 1.10 g tetramethylammonium chloride in 1 L distilled and deionized water), mix and wait 10 min, 4 vortex; 5Add 1 mL anhydrous pentane, 6 vortex and allowed to separate; 7 discard the organic (top) layer; 8 transfer aqueous (bottom) layer to a amber autosampler vial for HPLC-FLD analysis.	Flow: 1 mL/min	Time _{an} : 25 min, Time _{tot} : 30 min	LOQ: B ₁ 7.55, B ₂ 8.5 µg/L
(J. Wa ng, Zho u, and Wa ng 200 8)	B1		1 Add ACN/H ₂ O 1:1; 2 Shake over night; 3 Filter; 4Take 10 mL; 5 Place on the ice for 15 min; 6 Centrifuge to 7000 rpm, 10 min at 4 °C	Alltima C ₁₈ , 250 x 4.6 mm, 5 μ m 20 μ L of sample A) H ₂ O/TFA, B) ACN/FA 0-20% B from 0 to 5 min, 20-40% B from 5 to 10 min, 40-80% B from 10 to 15 min, 80% B from 15 to 20 min, 80-0% B from 20 to 25 min		ELSD 45°C of drift tube
	co rn pr od uc ts	10	Clean up: 1 Preconditione with 2 mL of MeOH, 2 Transfer 50 mL of sample; 3 Apply to centrifugal tube with 300 mg of amberlite XAD-4; 4 Stir for 5 h; 5 Wash with 40 mL with deionized H ₂ O; 6 Elute with 3 mL MeOH; 7 Collect 8 Dry 65 °C, 9,- Reconstitute			· · · · · · · · · · · · · · · · · · ·

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			Flow: 1 mL/min	Analysis Time: 25 min	LOD/LO Q: 3000 µg/L
(ACN) Acetonitrile, (AcOH) Acetic acid, (Matrix Solid Phase Dispertion, (NR) Not re	AE) Appearance energy, (AmAc): amn eported, (PBS) Phosphate Buffer Soluti	nonium acetate, (AmFo) Ammonium formate, on, (PLE) Pressurize Liquid Extraction, (RT) column conditioning.	, (FA) Formic acid, (Hex) Room temperature, (t_{an}) and	Hexane, (MeOH) Methanol nalysis time, (t_{Tot}) total time	l, (MSPD) including

Table 4. Transitions for FBs						
FB ₁						
Transitions (m/z)	CE	DP	CoV			
722.2→704.3	31	70-76	50			
722.2→352.3	38-40	70-76	50-60			
722.2→334.3	38-56	70-76	50-65			
FB_2/FB_3						
706.2→354.4	37	68-75	50			
706.2→336.5	40-47	68-75	50-55			
706.2→318.4	40-55	68-75	50-65			
(CE) Collision Energy, (CoV) Cone Voltage, (DP) Declustering potential, all in V						

 Table 5. Non-chromatographic methods for detection of fumonisins

 Immunological methods
 Molecular methods

 Enzyme Linked Immuno Sorbent Assay (ELISA)
 Internal transcriber spacer (ITS)

 Dipstick
 Intergenic spacers (IGS)

 Biosensor
 Polyketide synthase

 Immunoaffinity
 FUM genes

 Colloidal gold immune assay
 Microarray

 Polymerase chain reaction (PCR)
 Polymerase chain reaction (PCR)

(Majdinasab, Aissa, and Marty 2021; Deepa and Sreenivasa 2019; Gong, Jiang, and Chen 2015; Mirón-Mérida, Gong, and Goycoolea 2021)

5. Remarks

Fumonisins are mycotoxins widely distributed in food products, mainly due to the contamination of cereals (such as bread, bread, pasta, boxed cereals, flour, among others) by species of the Fusarium genus. Additionally, their presence in livestock feed, along with the eventual accumulation of these mycotoxins within their tissues, increases the transmission chain. Fumonisins have a high capacity to withstand the processes used in the food industry. They have been found to be thermically stable, at a neutral pH, in temperatures ranging from 100-125°C, only observing small degrees of degradation in alkaline or acidic mediums at temperatures above 175°C. The analysis of the stability of its hydrolyzed forms in corn-based products indicates that their decomposition begins at temperatures above 250°C, with the loss of the TCA groups. Even so, their decomposition does not exceed 20% of total fumonisins. The conjugation of fumonisins with sugars, proteins and even metals also occur in food products that are rich in these chemical entities. Currently there are no specific methodologies for analysis, detection, and quantification of hydrolyzed or conjugated forms of fumonisins for all the interest matrices, representing a niche of opportunity from an analytical and application point of view in the food industry. The basis of food in Mexico is corn; therefore, its population may be exposed to the consumption of Fusarium mycotoxins. Currently, there is a lack of legislation regulating the consumption of these compounds. The creation of new legislation is important to achieve adequate control and management of mycotoxin levels in food to ensure adequate food health in this regard. Supporting material can be consulted at https://www.sciencedirect.com/journal/arabian-journalof-chemistry and provides Tables 1-3 as excel file to facilitate the user experience by allowing the reader to sort by matrix, detector, LOD, LOQ, analysis time, etc.

6. Conclusion

Cheap, easy, and fast analytical methods for fumonisin detection are important worldwide, especially for countries where the content of these toxins is not regulated. Implementing regulation aids in the control of food products and contributes to food safety. Molecular, immunologic, and chromatographic methods can be used for fumonisin analysis. Molecular methods present the disadvantage of being only qualitative but widely used to identify fumonisin producer species. While immunologic and chromatographic methods can be utilized for both, qualitative and quantitative analysis. Immunologic methods are highly specific and useful for free fumonisins; however, these are not recommended for conjugated fumonisins. Immunological or molecular assays are still in development and could be a reasonable screening approach with final quantification being carried out by robust chromatographic methods, although some ELISA methods are commercially available. There are a wide variety of chromatographic methods. These are used for all kind of studies and applied to all kinds of samples because they can be coupled to different detectors. Chromatographic analysis of FBs can be qualitative or quantitative, another advantage is that they can analyze different FBs at the same time. Among the chromatographic methods, different sensibilities can be reached thus, although ELDS presents LOD very close to the maximum permissible levels it is still a viable option as a screening method. The use of mass spectrometry analyzers provides a high sensibility and is appropriate when analyzing samples of different origin. UPLC and HPLC methods are reported, as well as different analyzers. Very low limits of detection can be achieved. Sample pretreatment can be sufficient by extraction with an organic solvent or mixtures of ACN, MeOH and water, sometimes using weak acids. Similarly, these mixtures are employed in a gradient elution with 0.1-0.3% of acid, however clean-up is suggested for these mixtures. Chromatographic methods have a greater versatility regarding the combination of columns and detectors that are available, which is part of the reason these methods are still employed generating more sensitive, shorter, and reliable results. The information of the chromatographic methods for fumonisin analysis developed in the last 16 years has been included in this review. This paper will facilitate to the reader to consider the methodological aspects of a method to analytical success. Thus, the readers will be able to combine and adapt these aspects between methods to their own necessity.

Declarations

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Declarations of interest Authors declare that they do not have any conflict of interest.

Author contributions

Y.D.O.A. compiled all the methods parameters and wrote the first version of the manuscript,E.S.R. reviewed the medical and biological implications of fumonisins, M.Á.R.C. reviewed all methods parameters, M.Y.R. reviewed the general redaction of the complete manuscript.M.Á.R.C and M.Y.R. guided the complete work. All authors participated in the redaction of the manuscript.

Availability of data and material Not applicable

Code availability Not applicable

For the present review, ethics approval, consent to participate and consent for publication are not applicable.

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